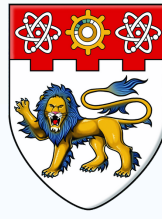




EARTH
OBSERVATORY
OF SINGAPORE



NANYANG
TECHNOLOGICAL
UNIVERSITY

Introduction to U-Th and ^{14}C dating

Hong-Wei Chiang (姜宏偉)

National Taiwan University

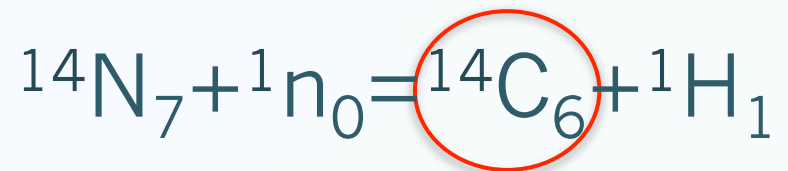
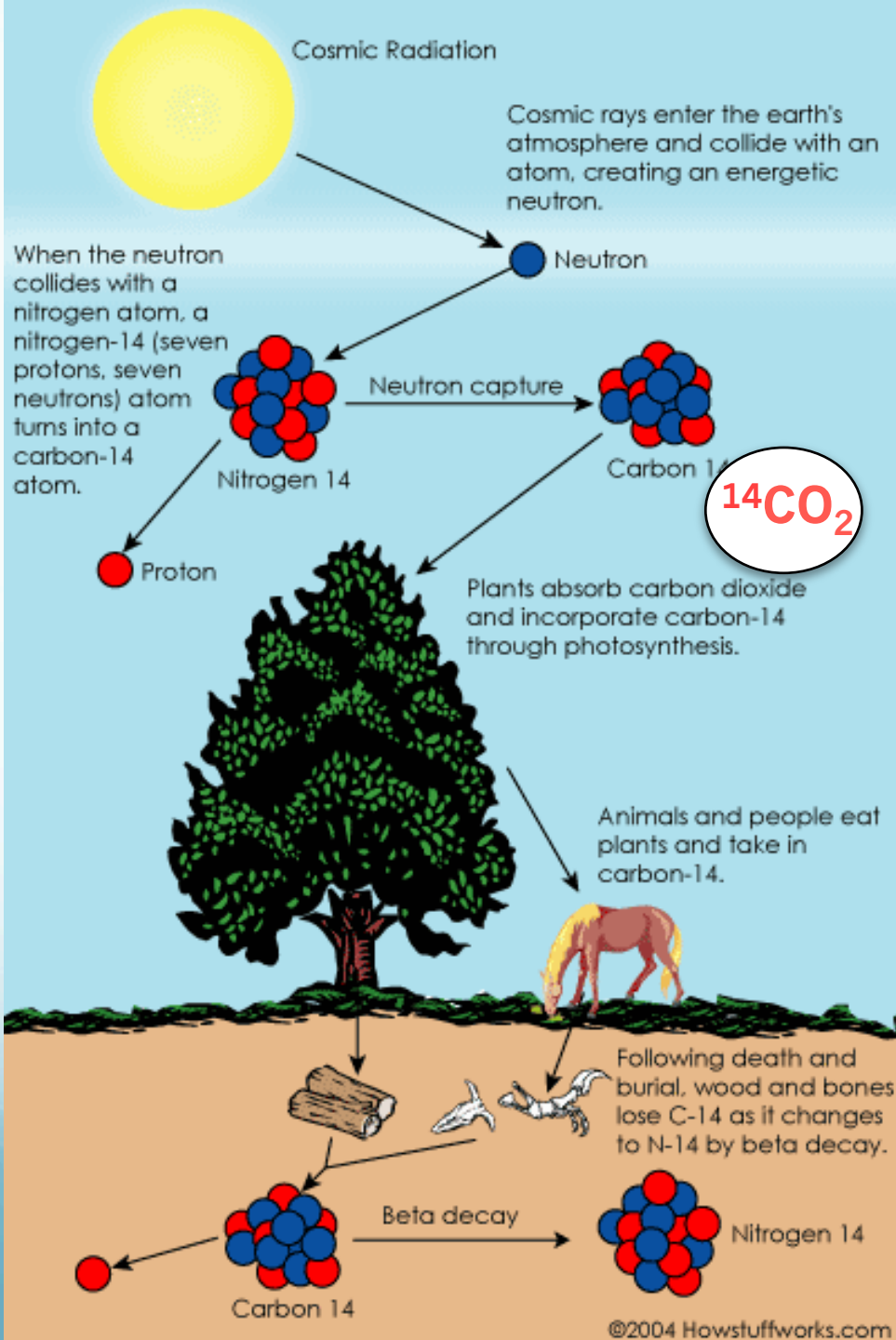
NCU 2016.12.30



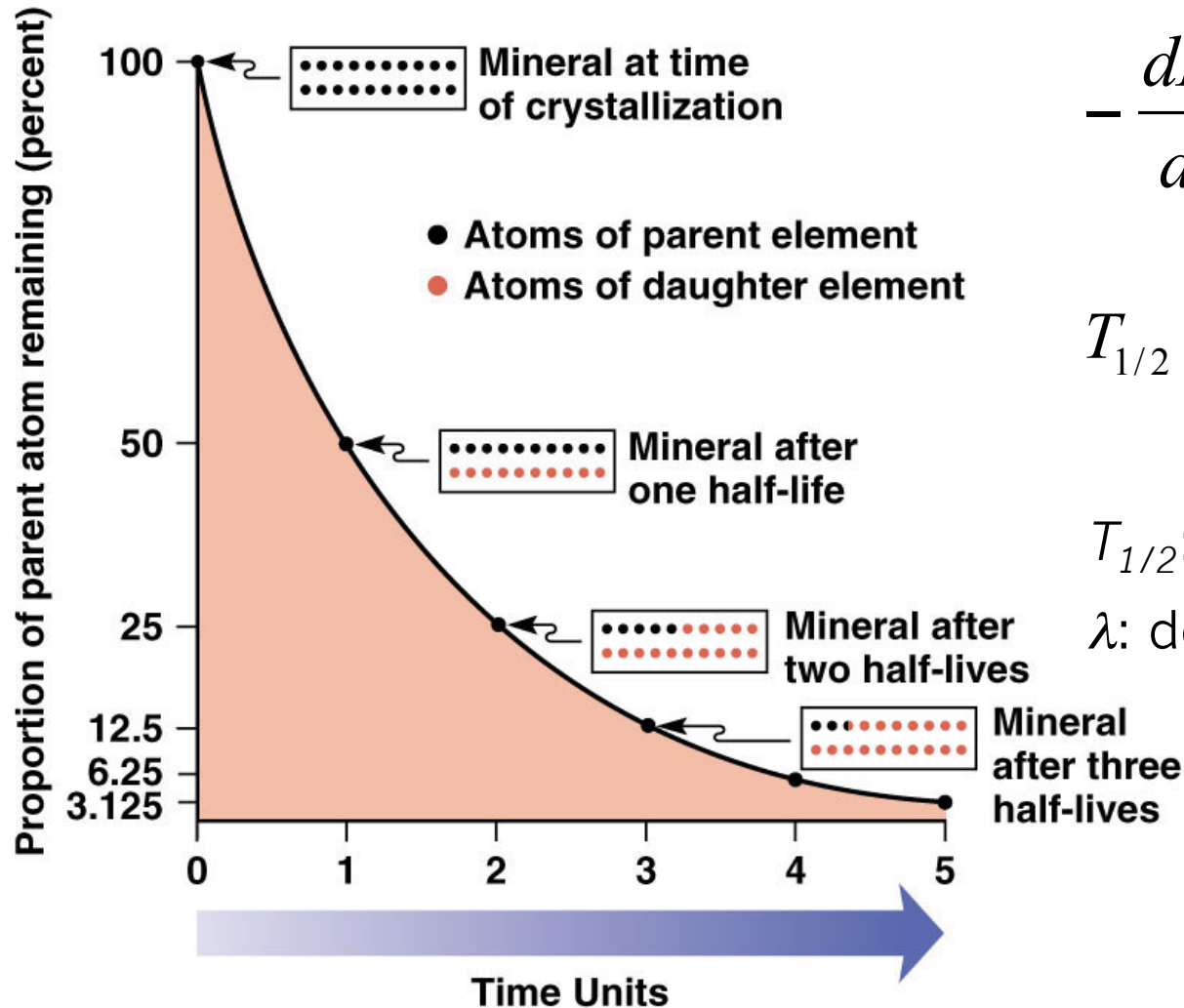
Professor Willard F. Libby of the University of Chicago received the Nobel Prize in Chemistry in 1960:

"for his method to use Carbon-14 for age determinations in archaeology, geology, geophysics, and other branches of science."

- Basic concept
 - Physical and chemical details
 - Principles
- Dating considerations
 - Atmospheric variation
 - Isotopic fractionation
 - Reservoir effects
 - Contamination
- Calibration
 - Terrestrial ^{14}C archives
 - Marine ^{14}C archives



Geometric Radioactive Decay



$$-\frac{dN}{dt} = \lambda N$$

$$T_{1/2} = \frac{\ln 2}{\lambda} = \frac{0.693}{\lambda}$$

$T_{1/2}$: half-life (**5730** yrs)

λ : decay constant

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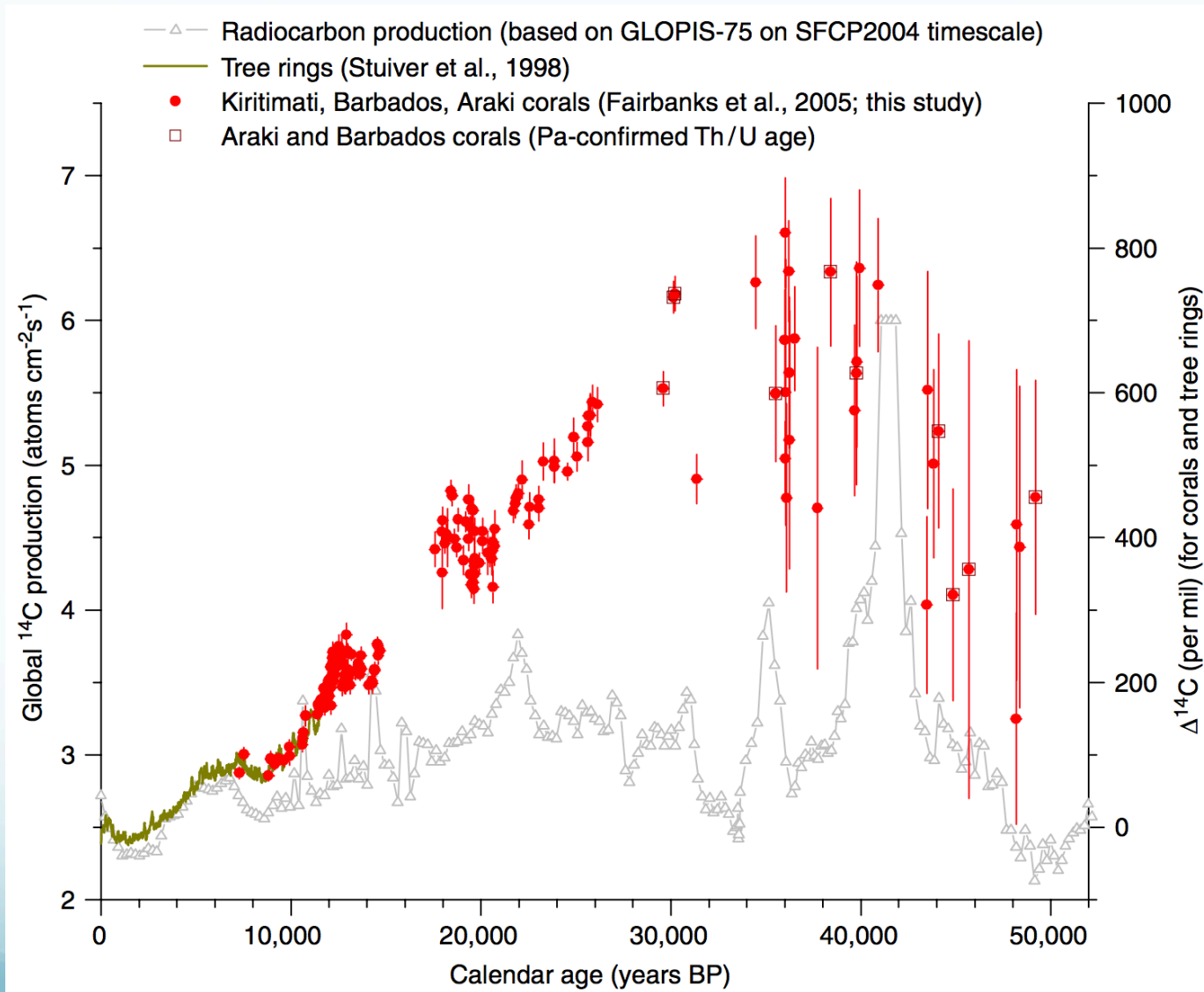
Atmospheric variation

Changed ^{14}C production over time

Suess effect

Bomb pulse

Changed ^{14}C production over time

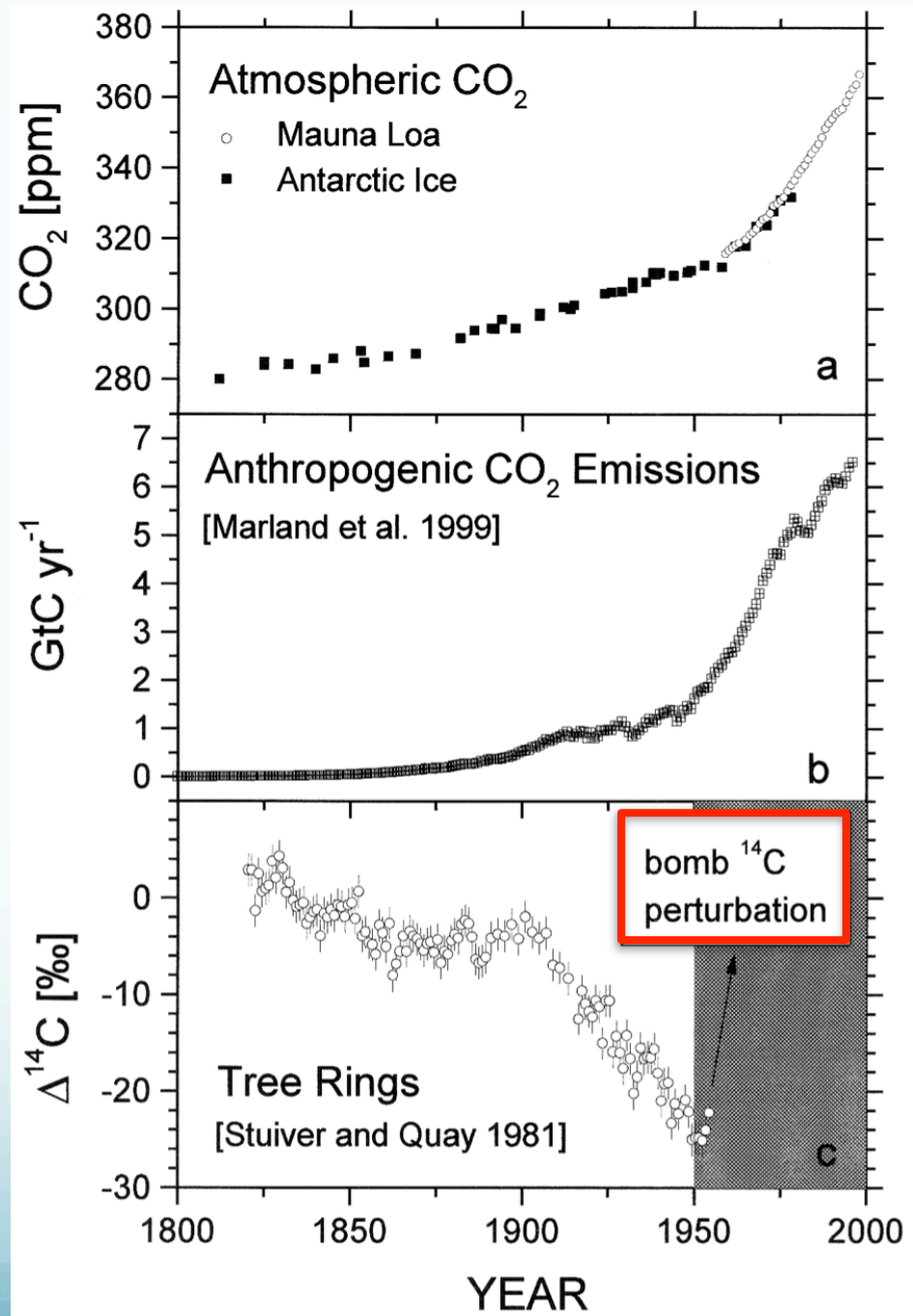


(Chiu et al., 2007)

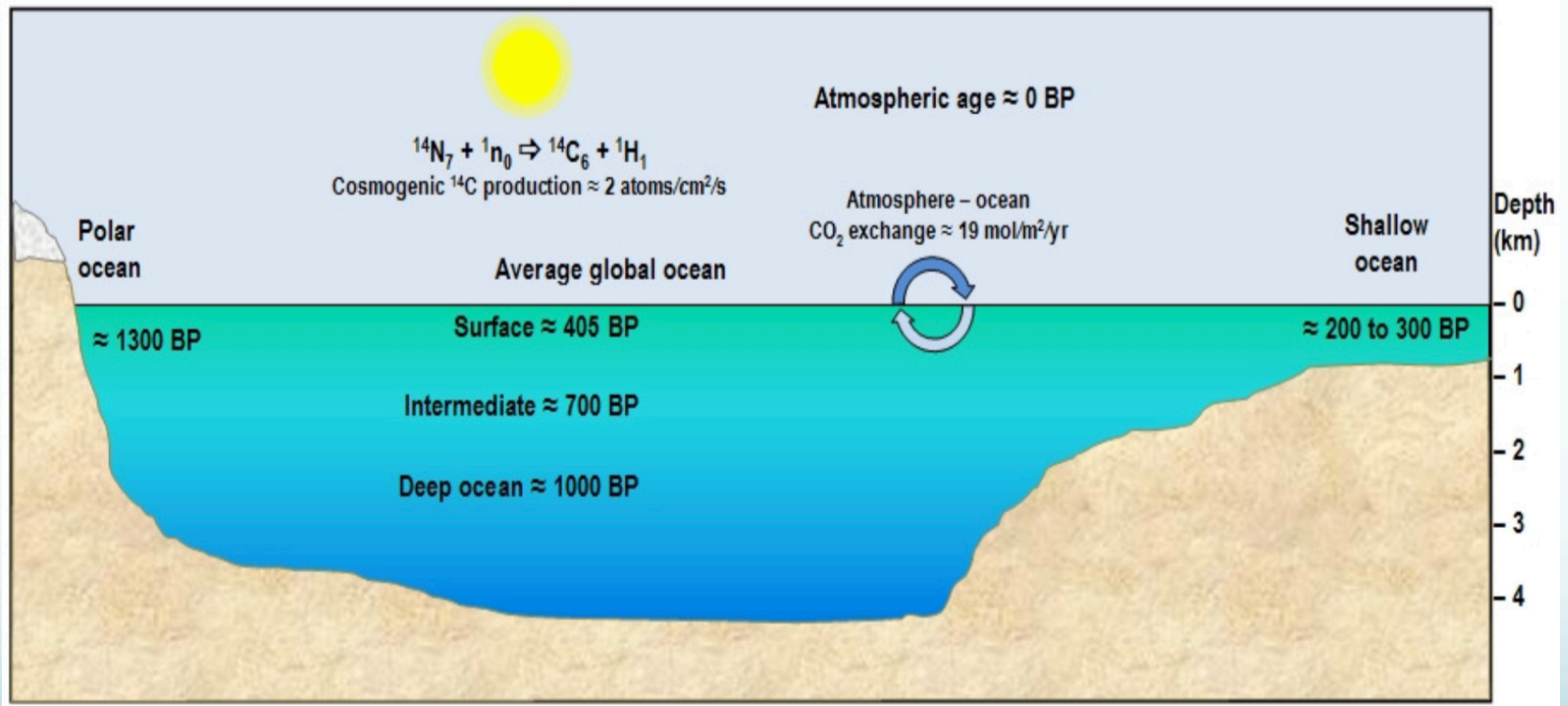
Suess effect

A change in the ratio of carbon isotopes (^{13}C and ^{14}C) in the atmosphere due to the admixture of large amounts of fossil-fuel derived CO_2 .

(Levin and Hesshaimer, 2000)



Reservoir effects

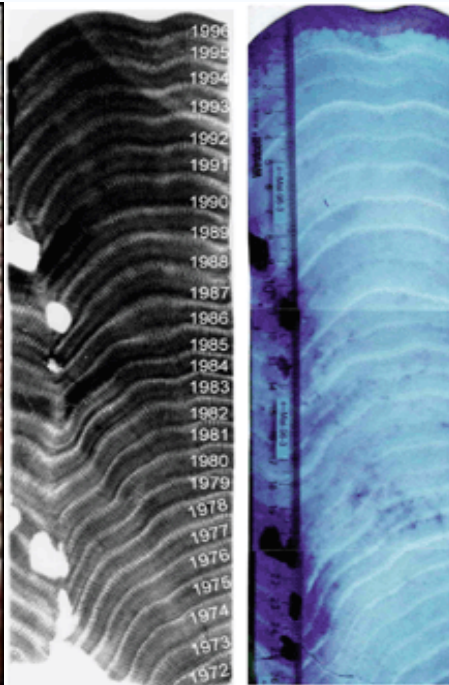


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Tree



Coral

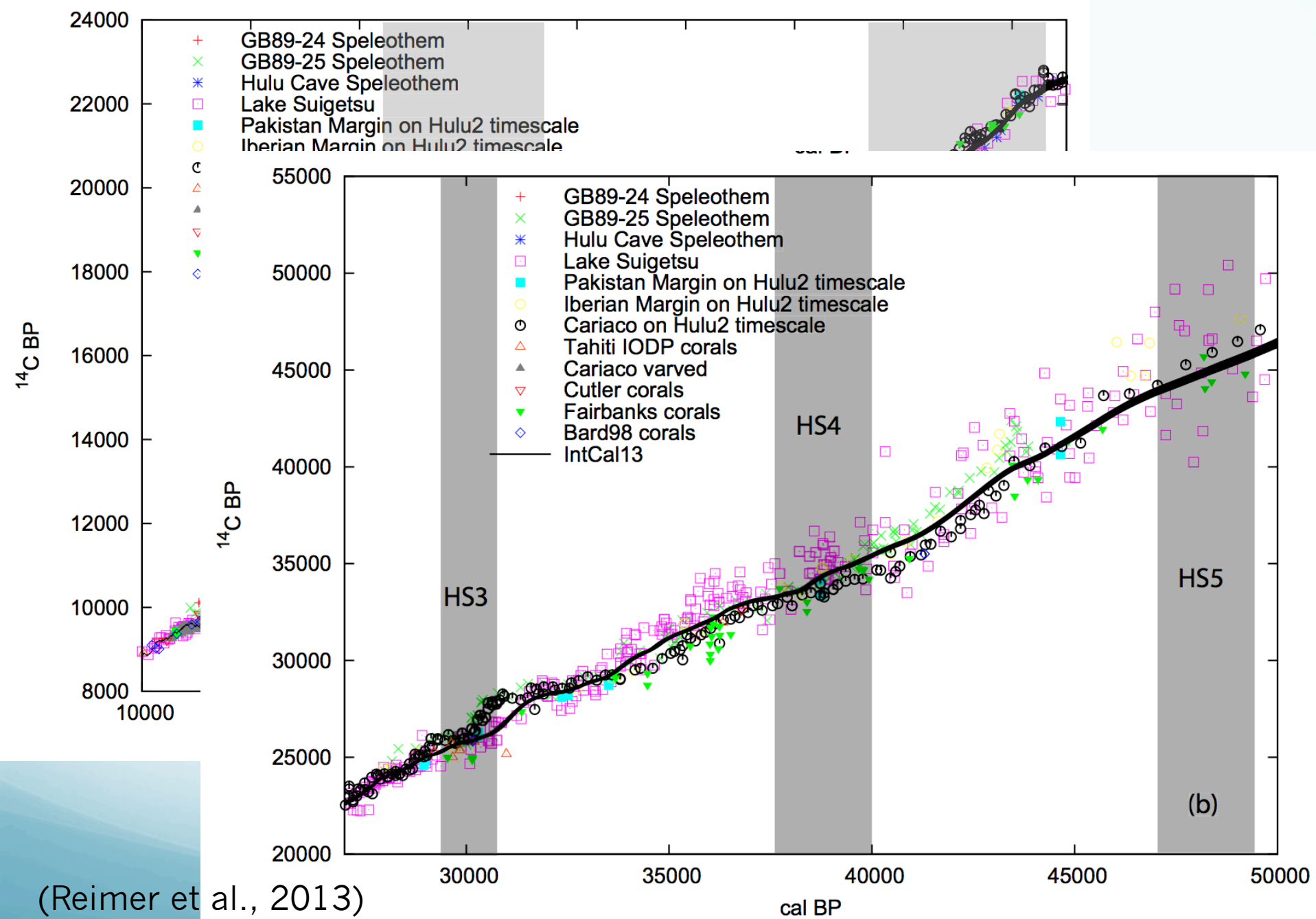


Speleothem



Sediments





■ Basic concept

Decay chains

Secular equilibrium and uranium-series dating

The ^{230}Th age equation

■ Test of dating assumptions

Initial mineralogy and petrology

Initial $^{230}\text{Th}/^{238}\text{U}$

Initial $^{234}\text{U}/^{238}\text{U}$

Initial U concentration

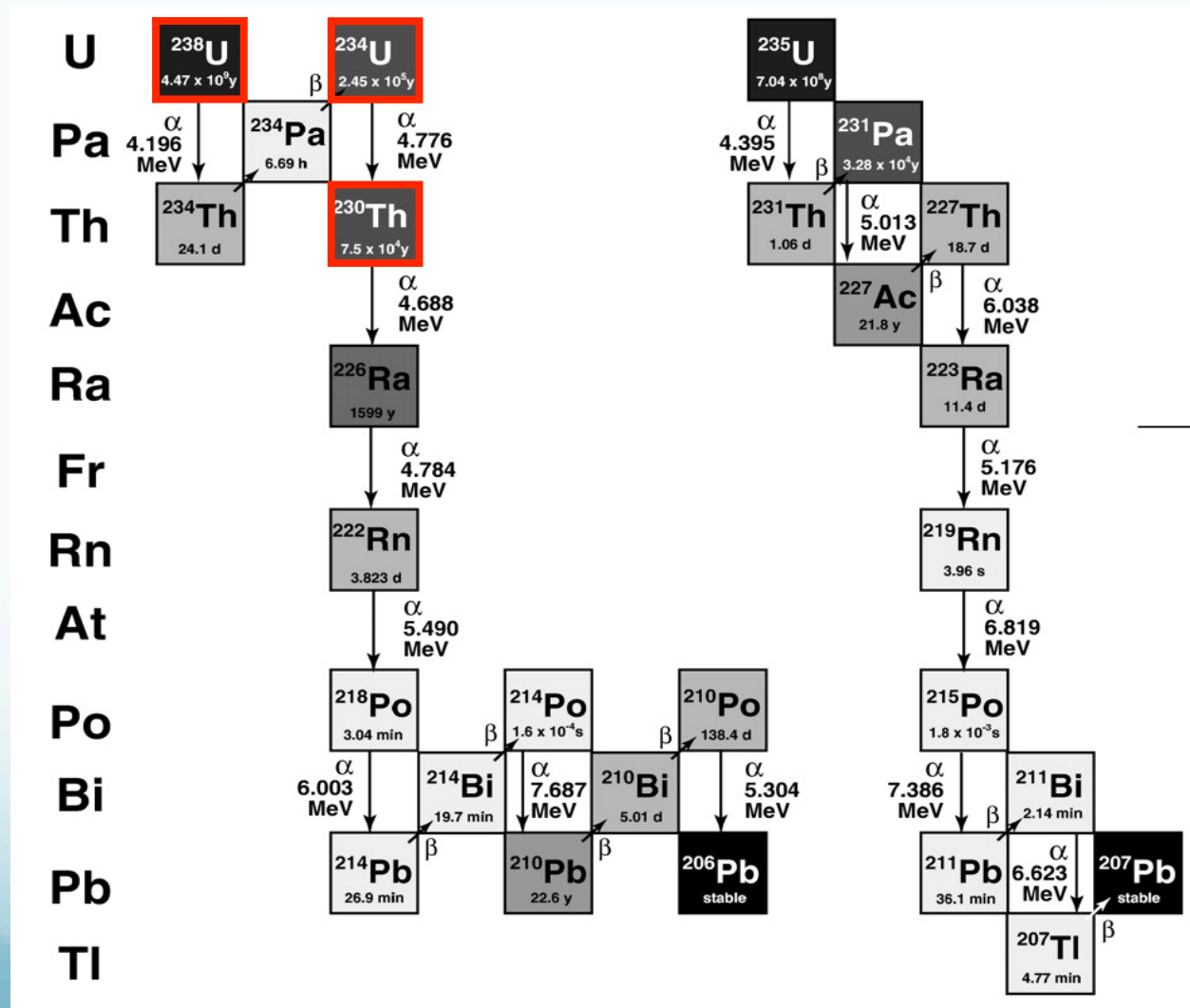
■ Sources of error in age

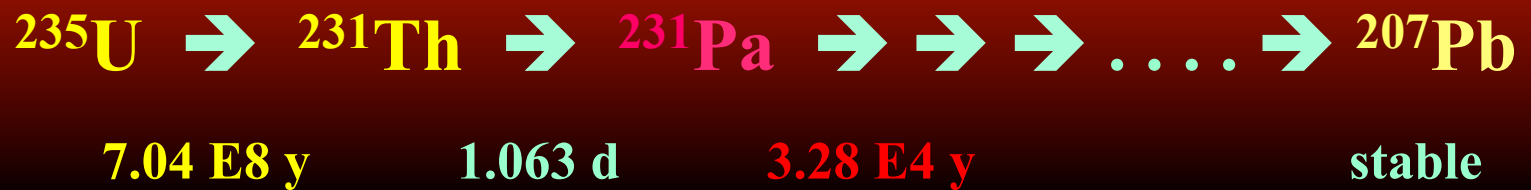
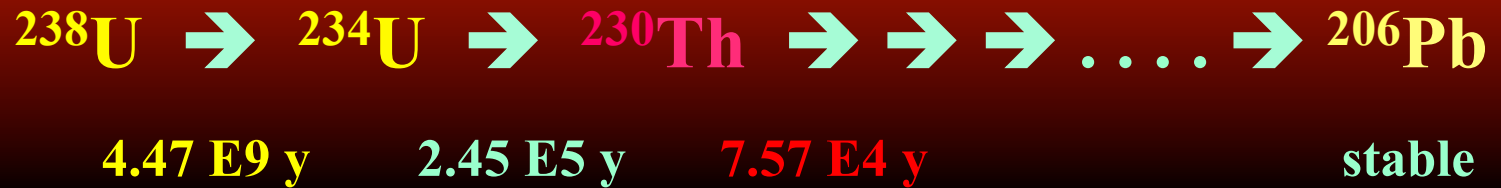
Error in measurement of isotope ratios

Error in half-lives and decay constants

Error in initial $^{230}\text{Th}/^{232}\text{Th}$

Schematic drawing of the ^{238}U and ^{235}U decay chains

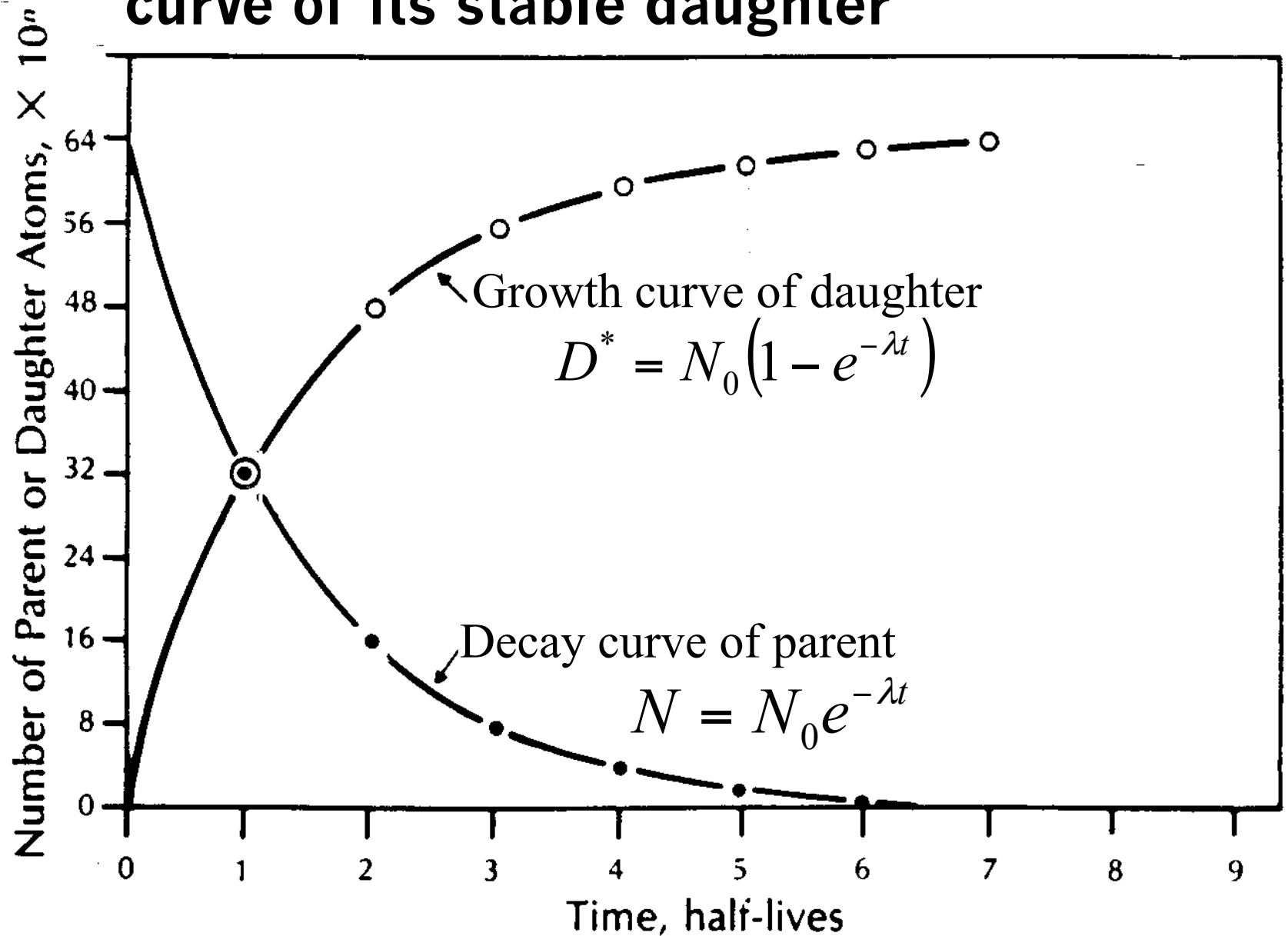




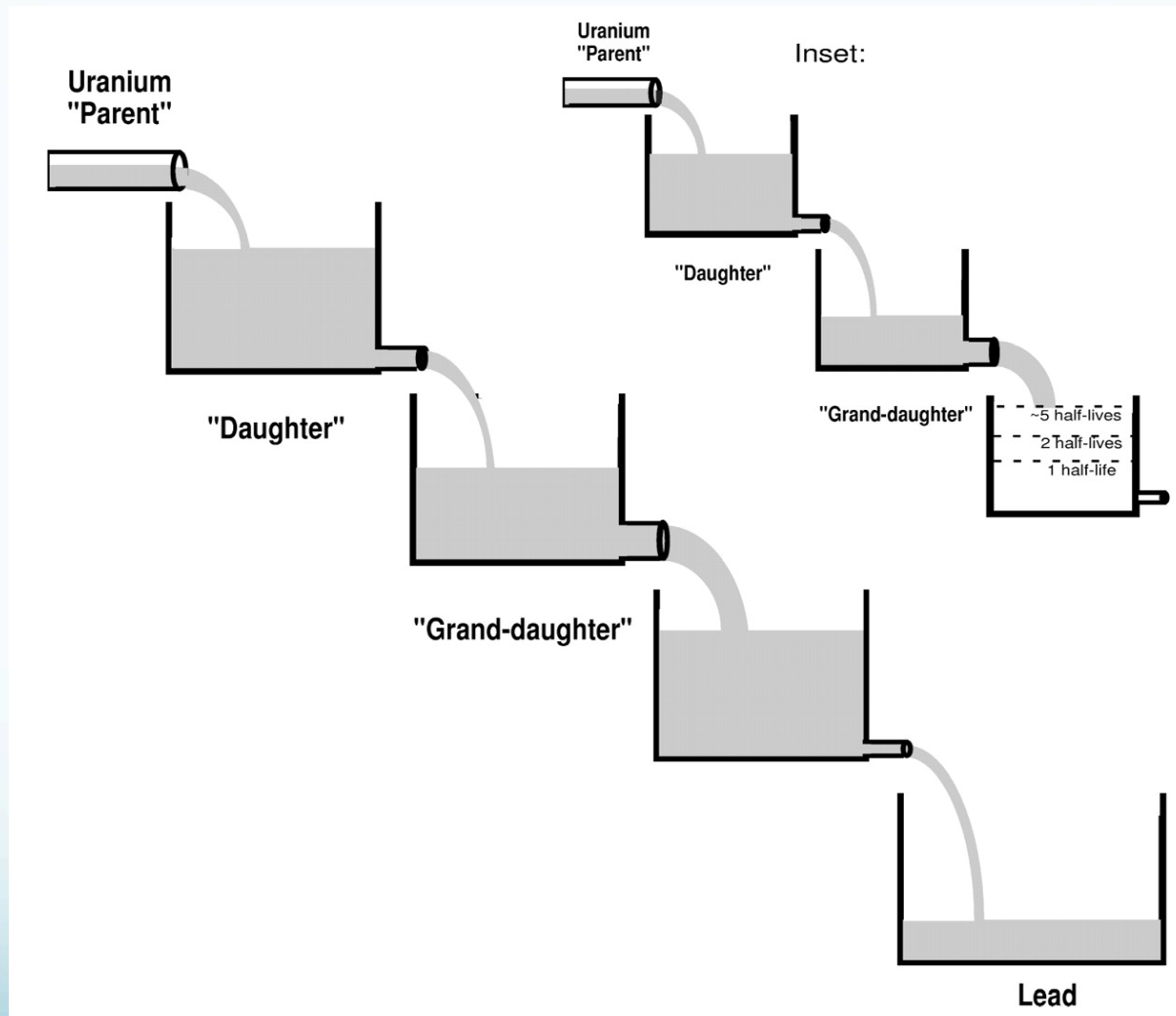
$$\frac{^{230}\text{Th}}{^{238}\text{U}} = 1 + \left(\left(\frac{^{232}\text{Th}}{^{238}\text{U}} \right) \left(\frac{^{230}\text{Th}}{^{232}\text{Th}} \right)^0 - 1 \right) e^{-\lambda_{230}t}$$

$$+ \frac{\delta^{234}\text{U}(0)}{1000} \left(\frac{\lambda_{230}}{\lambda_{230} - \lambda_{234}} \right) (1 - e^{(\lambda_{234} - \lambda_{230})t})$$

Decay curve of a radionuclide and growth curve of its stable daughter



At Secular Equilibrium $N_1\lambda_1 = N_2\lambda_2 = N_3\lambda_3 \dots\dots$



In contrast, we are using the disequilibrium of ^{238}U - ^{234}U - ^{230}Th .

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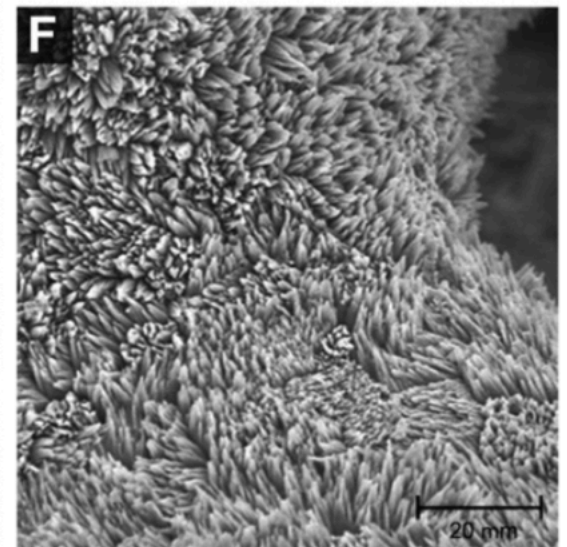
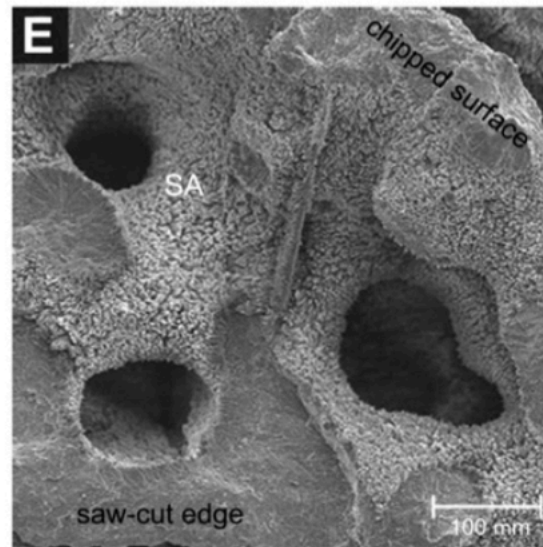
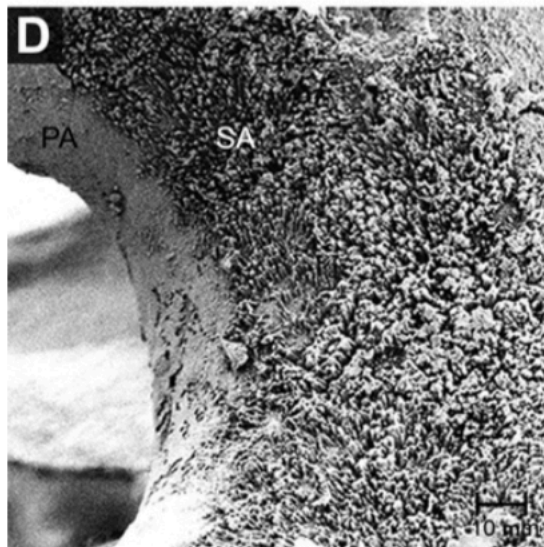
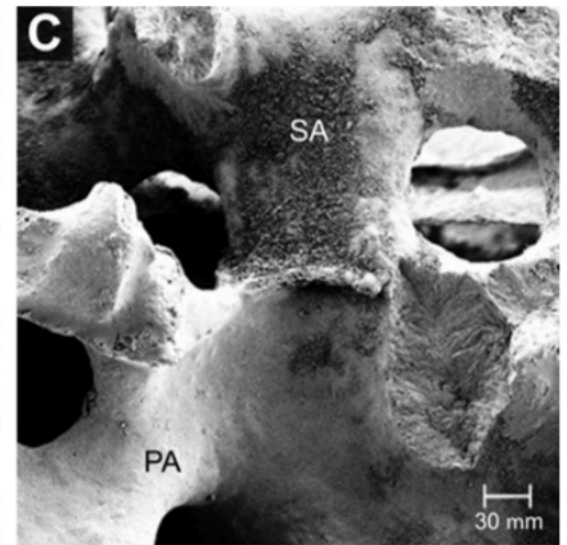
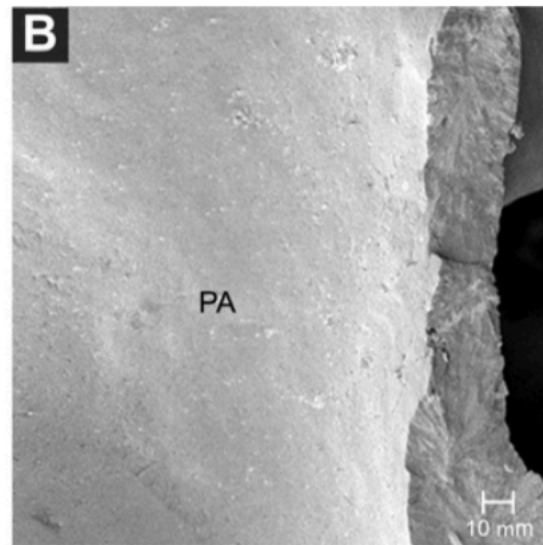
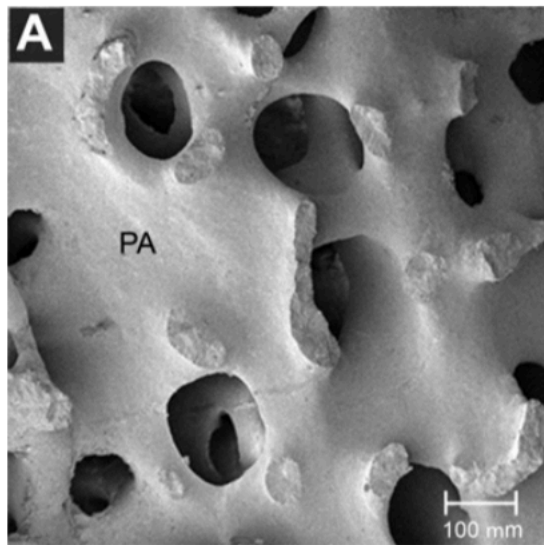
Initial U concentration

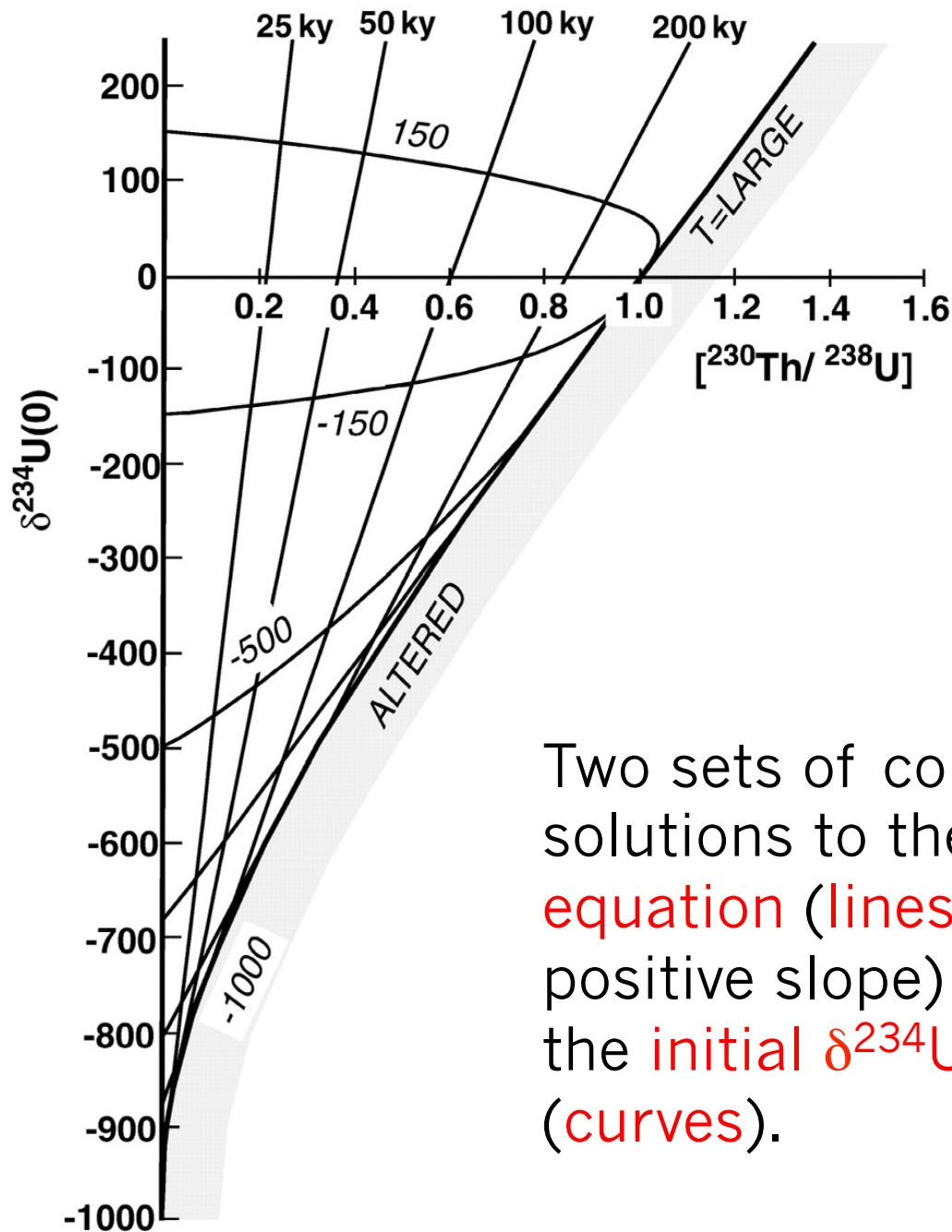
■ Sources of error in age

Error in measurement of isotope ratios

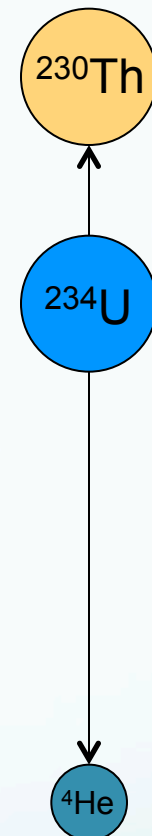
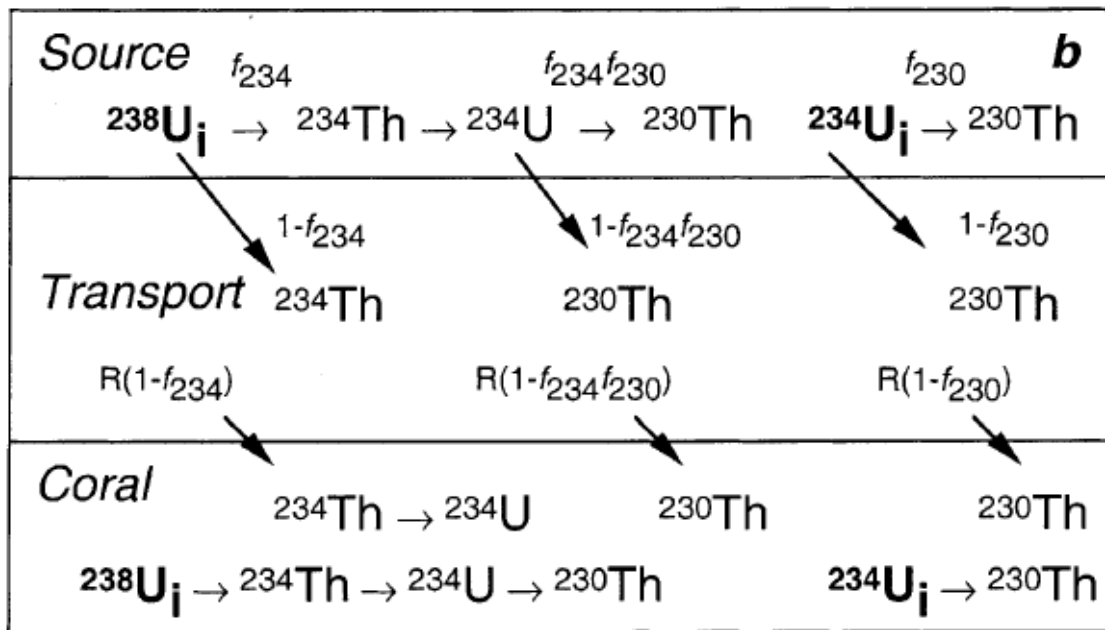
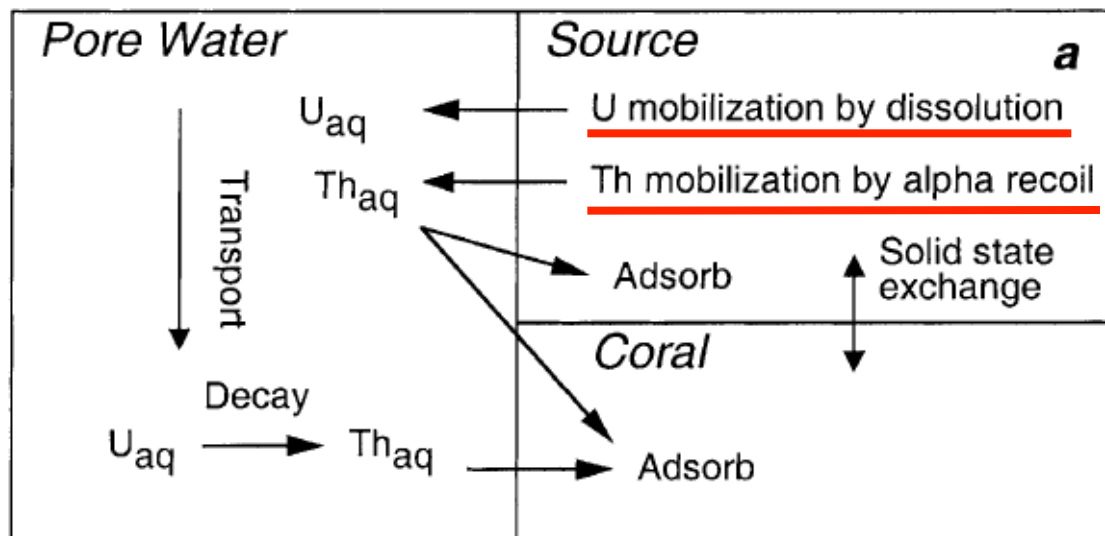
Error in half-lives and decay constants

Error in initial $^{230}\text{Th}/^{232}\text{Th}$





Two sets of contours indicate solutions to the ^{230}Th age equation (lines with steep positive slope) and solutions to the initial $\delta^{234}\text{U}$ equation (curves).



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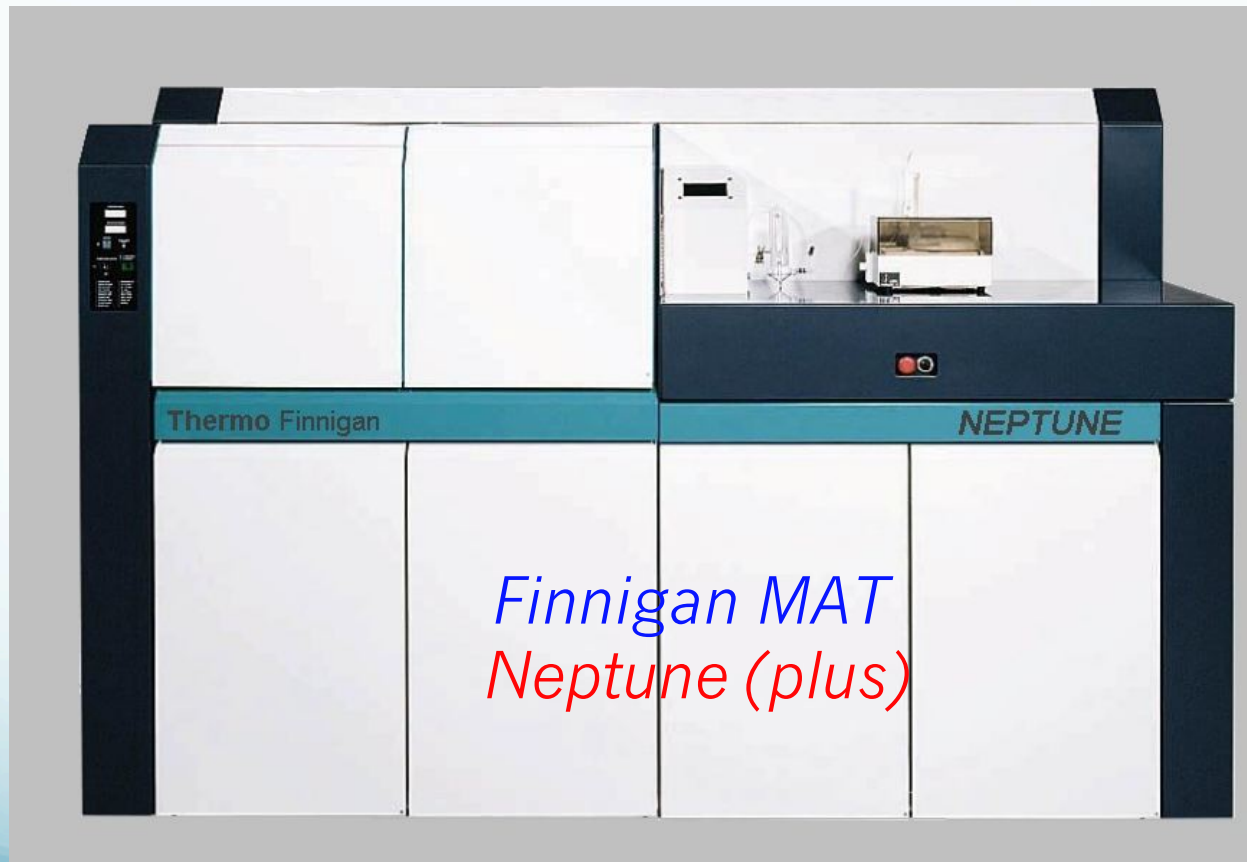
Initial U concentration

■ Sources of error in age

Error in measurement of isotope ratios

Multi-collector ICP-MS (MC-ICP-MS)

Higher efficiency of ionization and transmission due to better focusing



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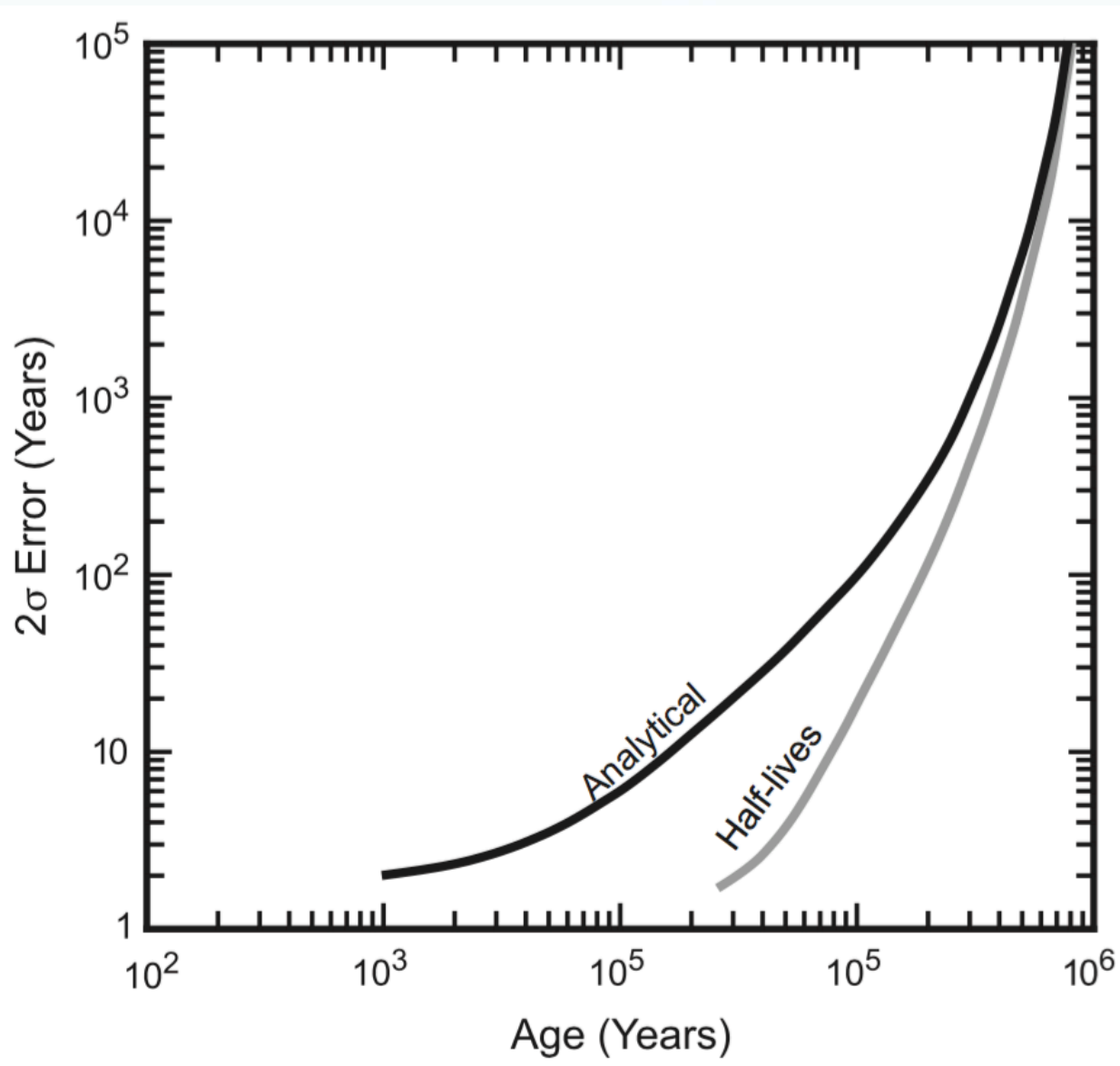
Initial $^{234}\text{U}/^{238}\text{U}$

Initial U concentration

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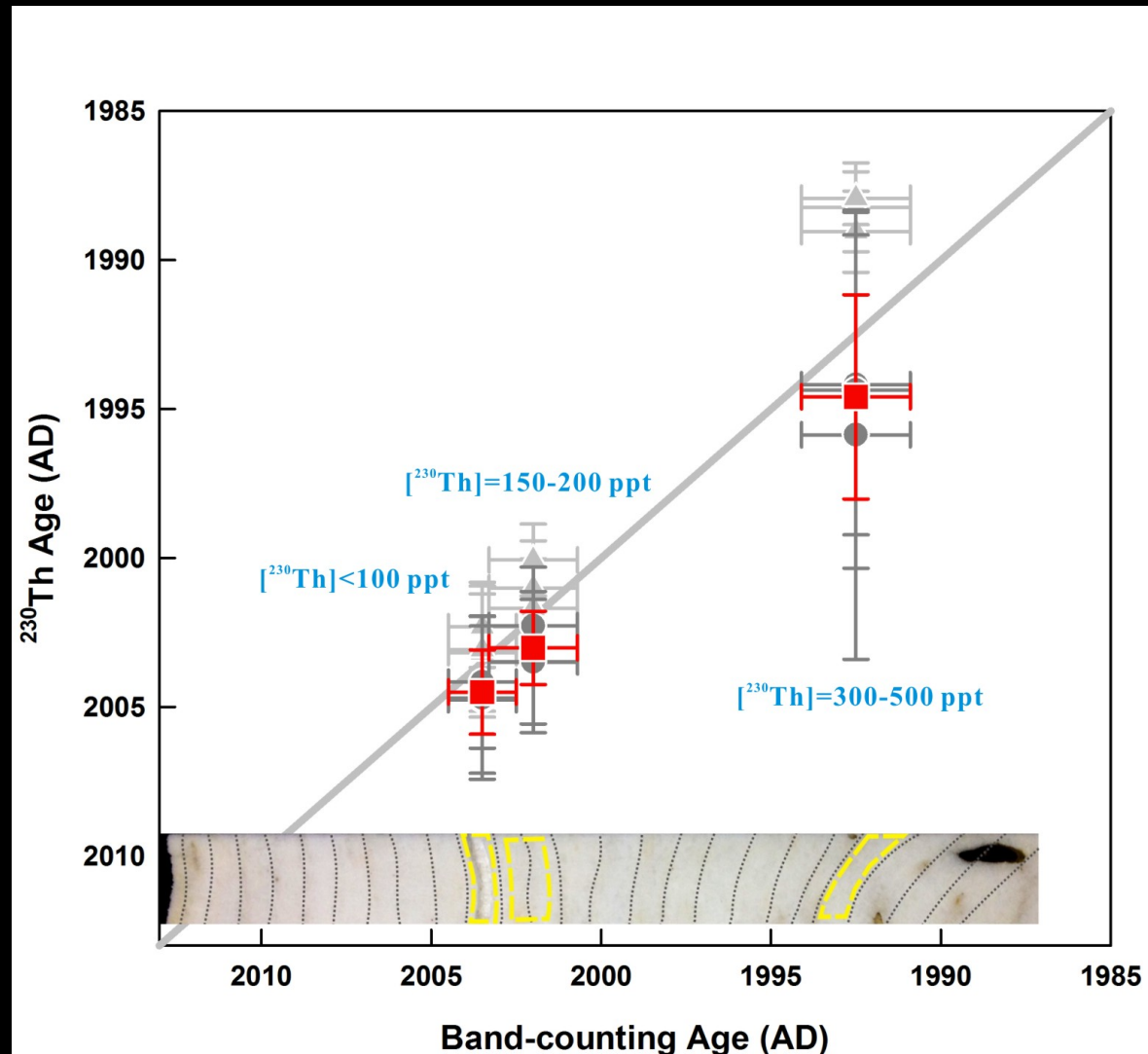
Problems

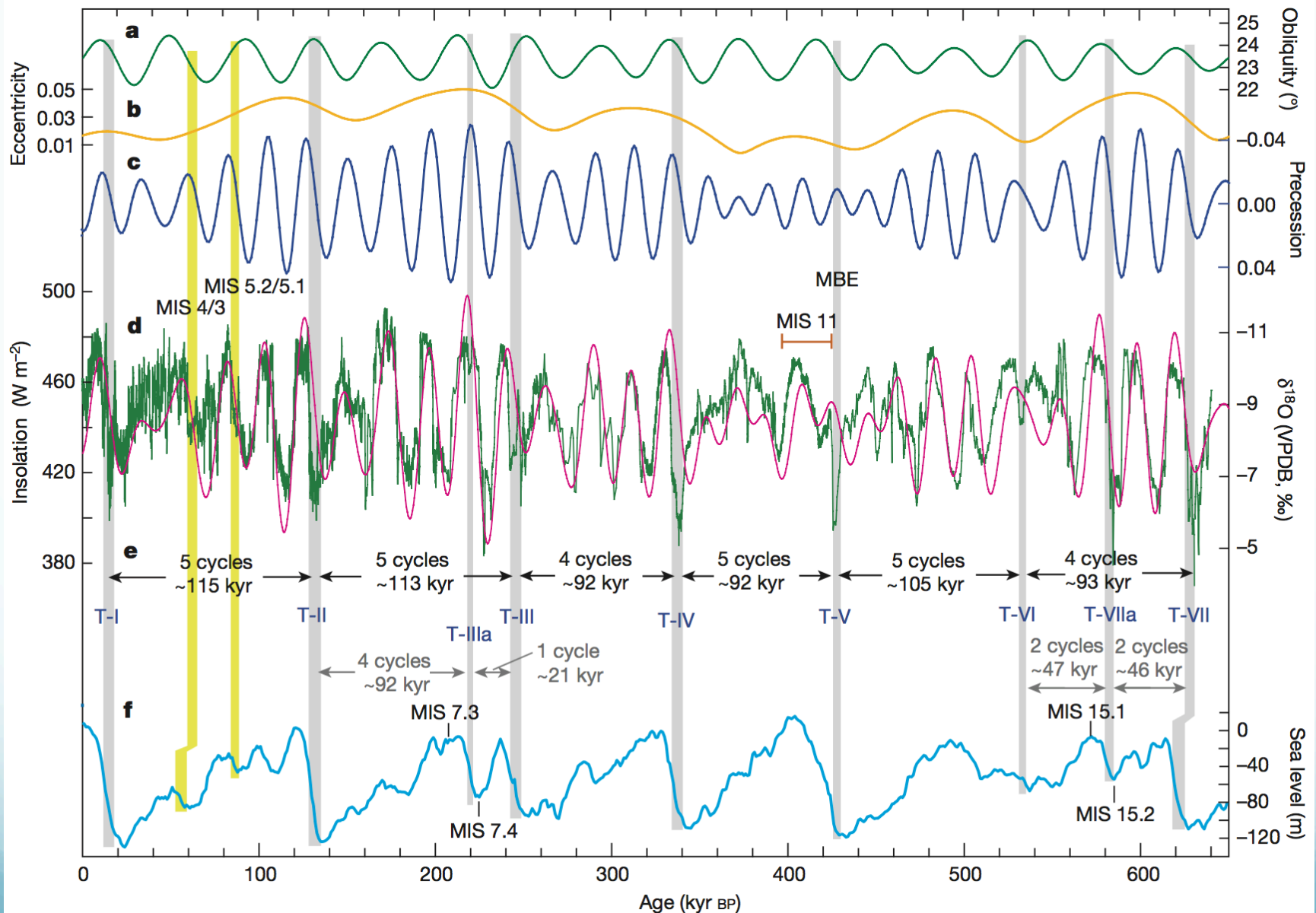
$$\frac{{}^{230}\text{Th}}{{}^{238}\text{U}} = 1 + \left(\left(\frac{{}^{232}\text{Th}}{{}^{238}\text{U}} \right) \left(\frac{{}^{230}\text{Th}}{{}^{232}\text{Th}} \right)^0 - 1 \right) e^{-\lambda_{230}t}$$
$$+ \frac{\delta^{234}\text{U}(0)}{1000} \left(\frac{\lambda_{230}}{\lambda_{230} - \lambda_{234}} \right) (1 - e^{(\lambda_{234} - \lambda_{230})t})$$

3. Initial ${}^{230}\text{Th}$

 6 years

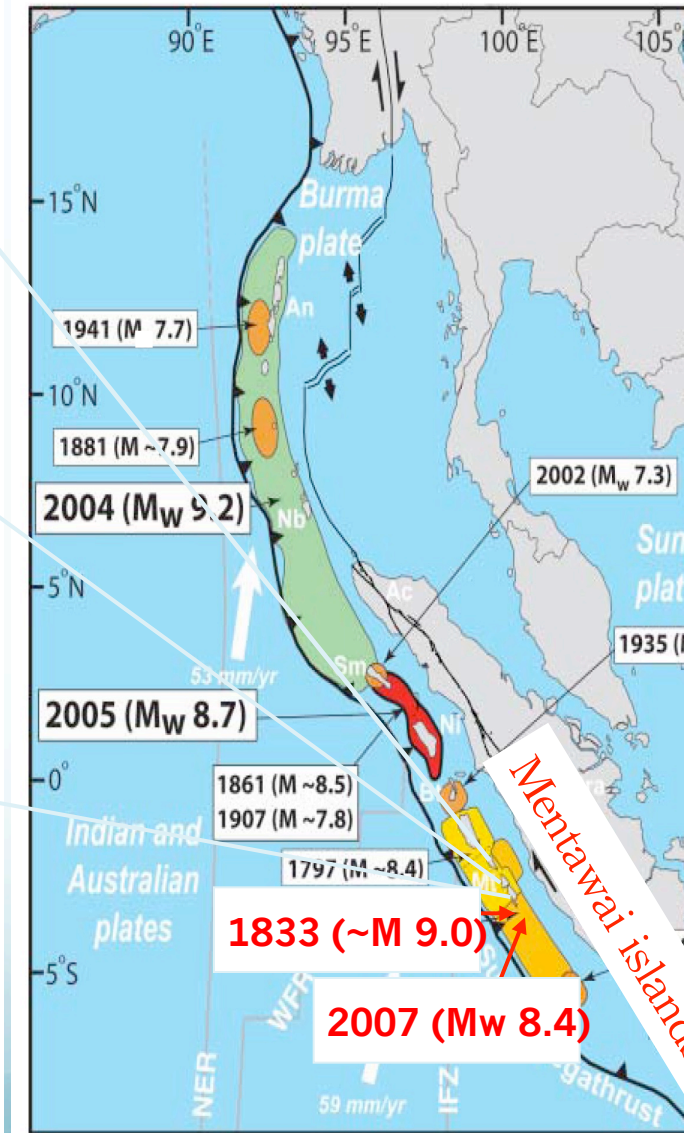
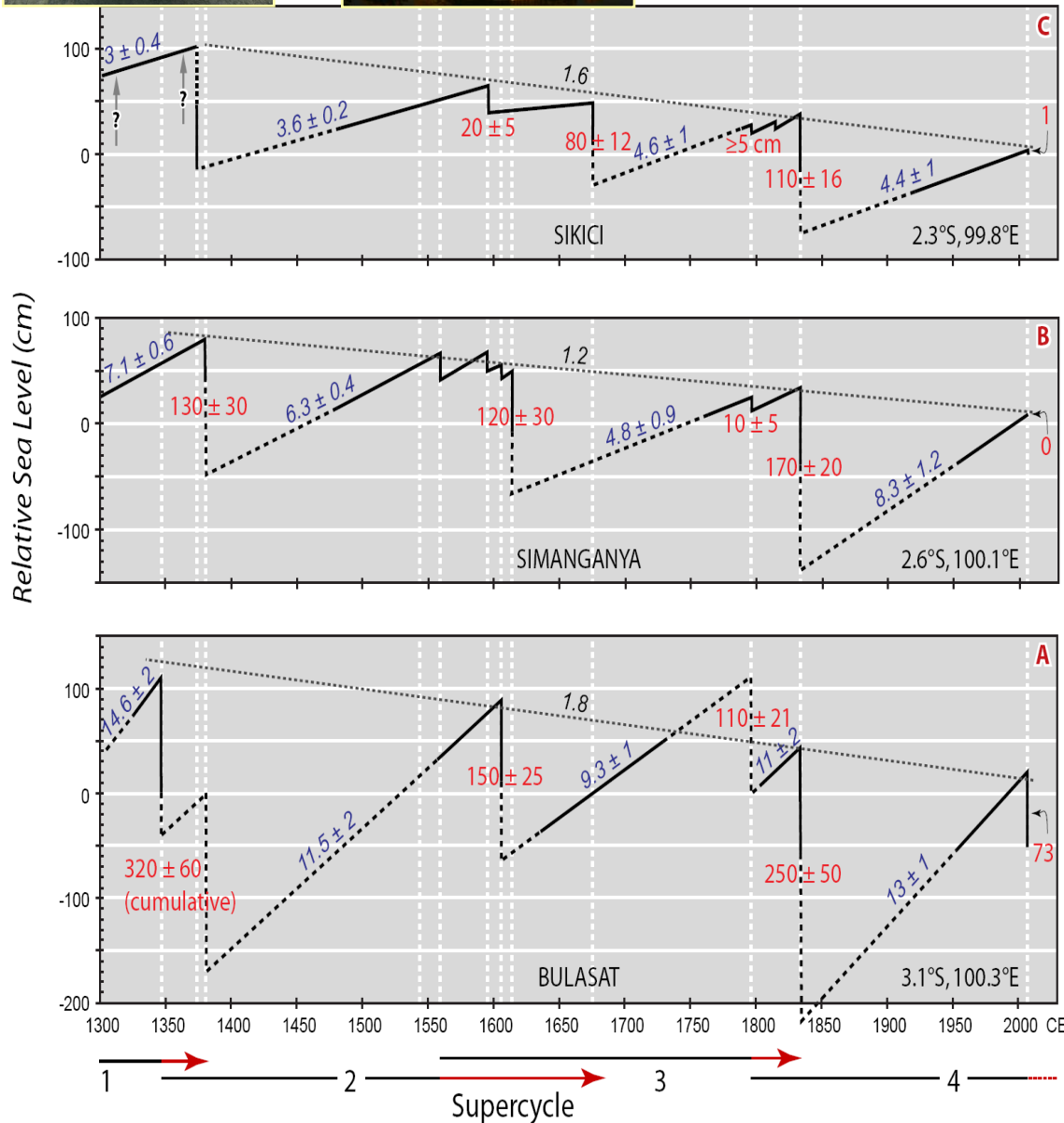
Capability of U-Th dating



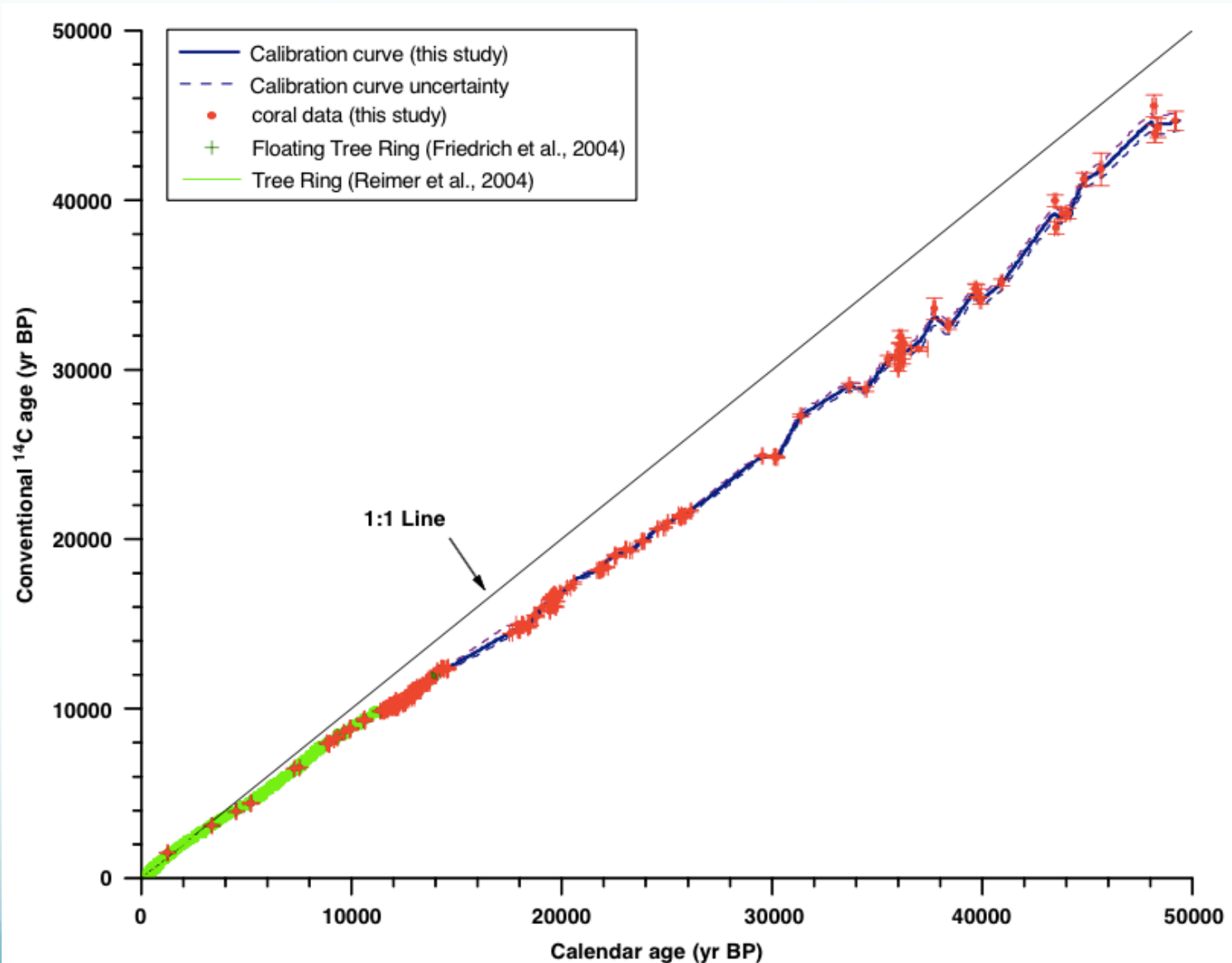




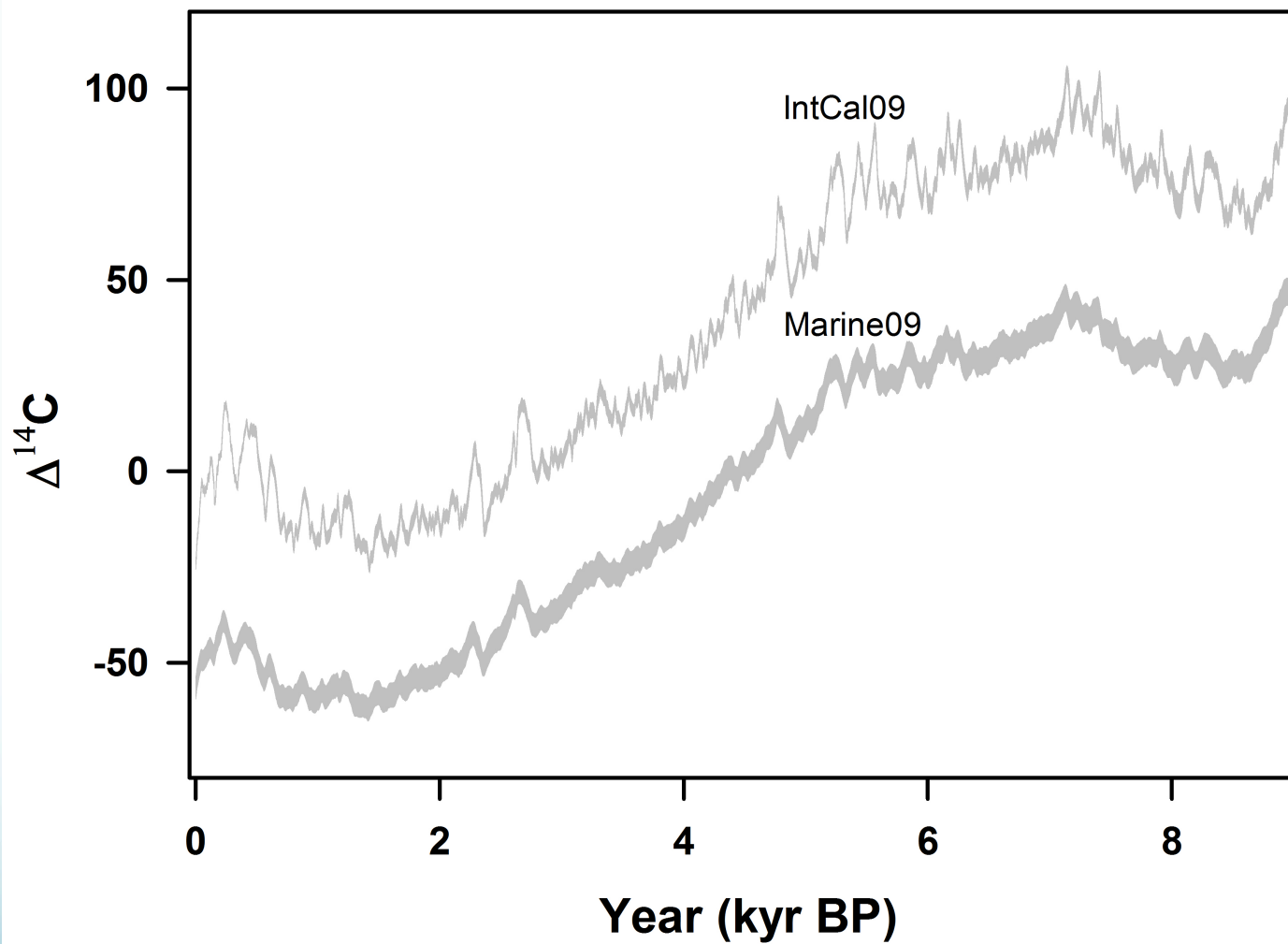
200-yr supercycles of giant earthquake clusters in the Mentawai islands Sieh et al., Science, 2008



Radiocarbon calibration



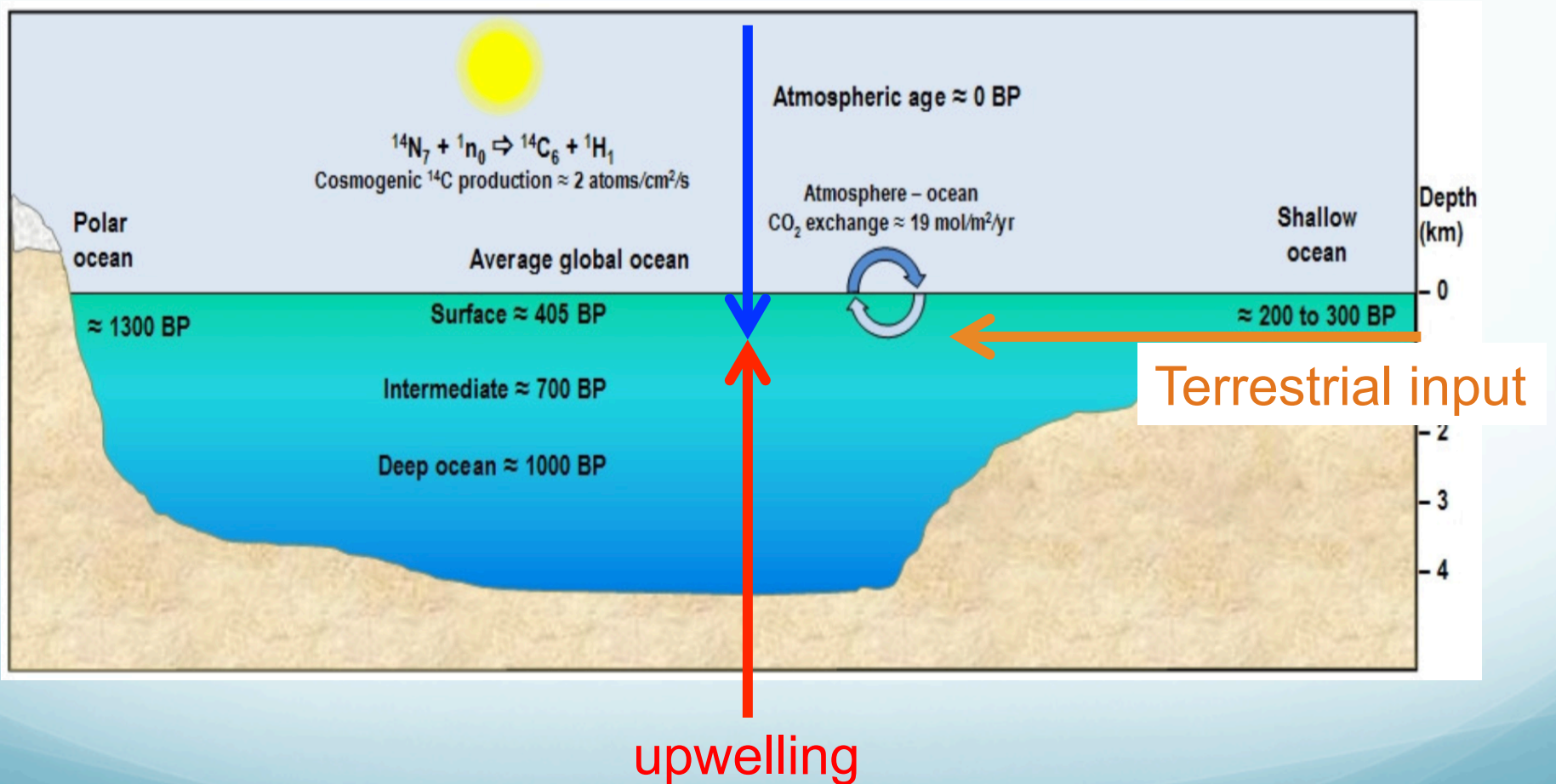
(Fairbanks et al., 2005)

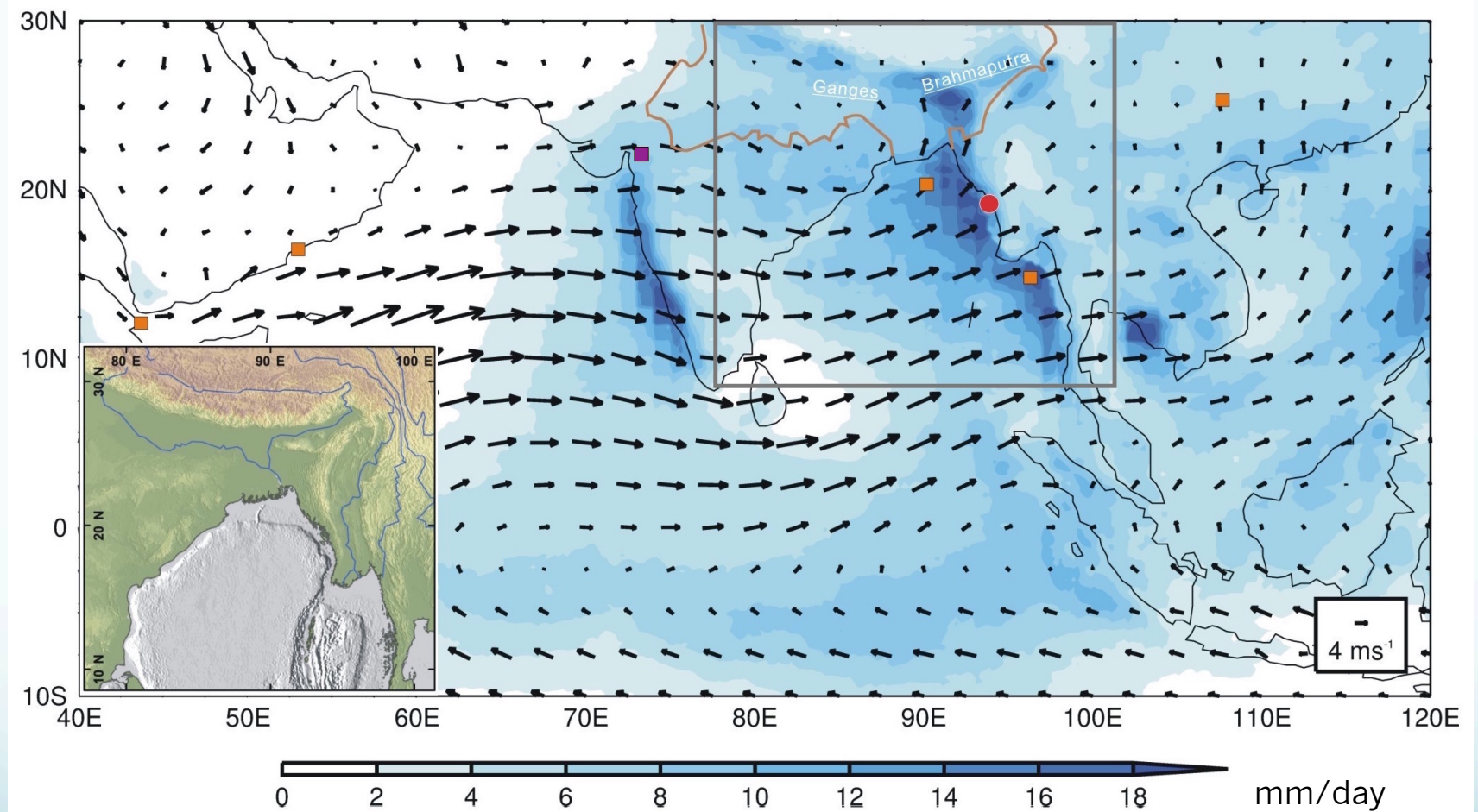


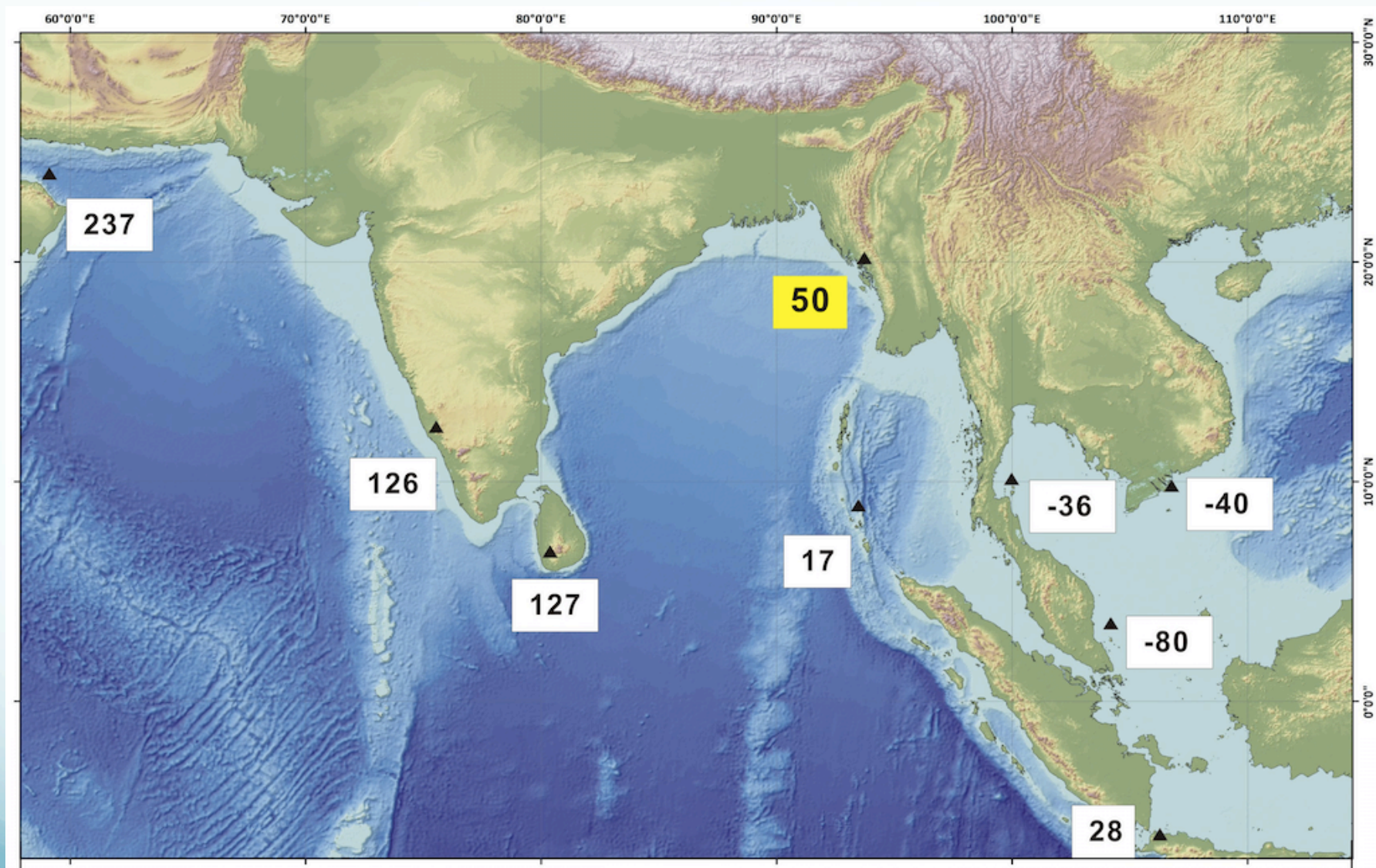
(Reimer et al., 2009)

ΔR distribution in ocean

Air-Sea interaction







Chiang et al., in preparation

Take-home message:

- ^{230}Th method can date samples as young as young 10-year olds with a precision of ~ 1 year.
- Use of a constant ΔR value to calibrate ^{14}C data may not be appropriate particularly in where has abundant terrigenous input.
- Time- and location-specific regional ΔR (R) value is strongly recommended for future ^{14}C dates.

Thanks for your attention

