



## Nitrogen as the carrier gas for helium emission along an active fault in NW Taiwan

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### ABSTRACT

Variations of He gas concentration are widely applied in studies devoted to the location of faults and to monitor seismic activities. Up to now, its migration mechanism in soil is not fully understood. A systematic soil gas survey across an active fault in NW Taiwan provides the opportunity to closely examine the mechanism of He migration in the fault zone. Significant spatial and temporal correlations observed between soil N<sub>2</sub> and He gas support the hypothesis that N<sub>2</sub> is the probable carrier gas for He emission in the studied area. Based on N<sub>2</sub>/Ar ratios and N<sub>2</sub> isotopic results, the excess soil N<sub>2</sub> in this study is considered to be largely derived from ancient atmospheric air which was dissolved in groundwater. Furthermore, observations rule out the possibility of CO<sub>2</sub> being the dominant carrier gas for He in the studied area based on the C and He isotopic compositions and the relationship between concentrations of these gases. At least two soil gas sources, A and B, can be identified in the studied area. Source A is an abiogenic gas source characterized by excess N<sub>2</sub> and He, and very low O<sub>2</sub> and CO<sub>2</sub> content. Source B, on the other hand, is a mixture of biogenic gas and atmospheric air. The development of the fault system is an important factor affecting the degree of mixture between sources A and B. Therefore, variations of soil gas composition, in particular those derived from source A, could be a useful proxy for tracing faults in the area.

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### 1. Introduction

Trace gas is one of the most useful tools which have been applied to discover and delineate faults (e.g. Ciotoli et al., 1998; Toutain and Baubron, 1999; Fu et al., 2005, 2008; Walia et al., 2005b, 2008), study/monitor seismic activities (e.g. Tsunogai and Wakita, 1995; Sano et al., 1998; Italiano et al., 2001; Gulec et al., 2002; Chyi et al., 2005; Walia et al., 2005c; Yang et al., 2005b, 2006b; Fu et al., 2009), study/monitor volcanic activities (e.g. Sano and Wakita, 1985; Hilton et al., 2000, 2002; Lee et al., 2008) and as an indicator for fluid sources (e.g. Tedesco and Scarsi, 1999; Van Soest et al., 2002; Jaffe et al., 2004; Chen et al., 2005; Walia et al., 2005a; Yang et al., 2003a, 2004, 2005a). Helium-4 (hereafter simply referred to as He) is important for these purposes, because of its unique characteristics of being chemically inert, of non-biogenic origin, and being highly mobile, relatively insoluble in water, and radioactively stable (Reimer, 1980; Ozima and Podosek, 2002; Yang et al., 2009). Despite its importance, its migration mechanism is still debated.

In this study, a soil gas survey across the Hsincheng Fault (Fig. 1a and b), which is an active fault in NW Taiwan, was con-

ducted. Various gas concentrations (He, Ar, O<sub>2</sub>, CO<sub>2</sub>, N<sub>2</sub>) and isotopic compositions (<sup>3</sup>He/<sup>4</sup>He, δ<sup>13</sup>C<sub>CO2</sub> and δ<sup>15</sup>N<sub>N2</sub>) were determined in order to understand the origin of the soil gases. Weekly monitoring of soil gas concentrations was likewise conducted at a selection of the stations in order to understand the temporal variation of these gas species. The goal of this work is to contribute some observations pertaining to the mechanism of He gas transport. The carrier gas mechanism was considered in this case for He transport (Kristiansson and Malmqvist, 1982). The most likely carrier gas for He, either CO<sub>2</sub> or N<sub>2</sub>, which are the dominant soil gases in the study area, will be identified. Methane will not be considered here because it occurs only as a minor component in only three samples. To identify the possible carrier gas of He, the following aspects were studied: (1) the spatial variation of different gas species across the fault zone; (2) the temporal variation of various gases; (3) the origin of the different gases.

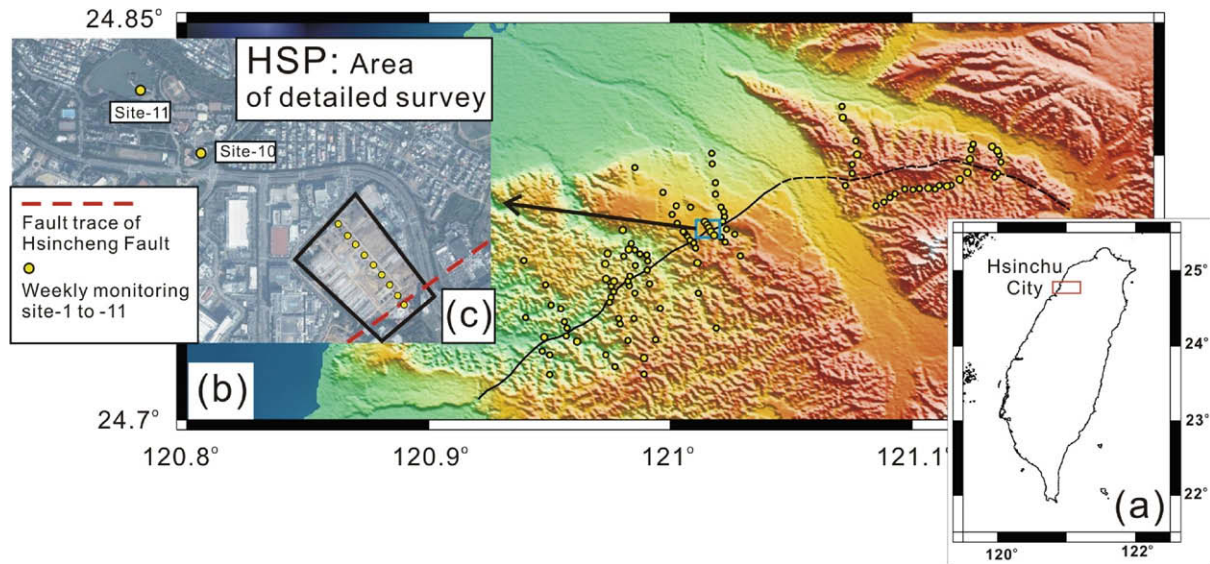
### 2. Principle and methodology

#### 2.1. Principle

Helium has two naturally-occurring radiogenic isotopes: <sup>3</sup>He and <sup>4</sup>He. Most of the <sup>4</sup>He is of radiogenic origin, i.e., the product of alpha decay of radioactive isotopes such as <sup>238</sup>U, <sup>235</sup>U and

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**Fig. 1.** (a) Location of Hsinchu County in Taiwan. (b) The enlarged map of the rectangle in (a) showing the general survey area and sampling points. The solid line indicates the fault trace of the Hsincheng Fault while the dashed line indicates the suspected extension of the Hsincheng Fault. (c) The enlarged map of the rectangle in (b) showing Hsinchu Science Park (HSP) (background image was modified from Google Earth) and the monitoring sites inside (dots inside rectangle showing site-1 to -9; dots outside the rectangle showing sites -10 and -11). Fault trace is shown by the thick-dashed line.

$^{232}\text{Th}$  in the crust. Most of  $^3\text{He}$  is of primordial origin (Butt et al., 2000; Ozima and Podosek, 2002). Radon ( $^{222}\text{Rn}$ ), which is often used in fault-finding studies (e.g. Fu et al., 2005; Yang, 2008), is produced in the radioactive decay series of  $^{238}\text{U}$ . The low diffusion coefficient of He and the short half-life of Rn (3.82 days) do not allow them to migrate to the surface only by diffusion. As a result, the mechanism by which such trace gases (i.e. He and Rn) migrate in the subsurface is still not clear.

Several different mechanisms have been proposed to explain the behavior of trace gas transport. Diffusion has long been considered as an important process for earth degassing (Newton and Round, 1961). However, many problems arise if long-distance transport is considered due to the small diffusion coefficients of trace gases. Helium, for example, has a diffusion coefficient in limestone or saturated rocks in the range from  $10^{-4}$  to  $10^{-9}$   $\text{cm}^2/\text{s}$  (Pandey et al., 1974; Lerman, 1979). It could only move a few tens of meters through solid rocks even at geological time scales if diffusion were the only mechanism (Etheridge et al., 1984; Gold and Soter, 1984). The other possible mechanisms are advection and groundwater transport. Advection refers to movement of materials under external forces such as pressure gradients. Such a mechanism requires a stream of free gas (a gas domain) which occurs only at sufficiently high concentration (Etiopie and Martinelli, 2002). However, trace gases like He usually do not occur in sufficiently high amounts to form a free gas phase. Moreover, such advection/convection flow is usually related to seismic or volcanic activities (King, 1978; Cox, 1980) and hence may not be applicable to the general situation. Newton and Round (1961) suggested that a significant portion of He is dissolved in water so that the migration of He is dominated by water flow. Such a model was supported by some subsequent studies (Eremeev et al., 1973). However, a groundwater transport model can only explain the long-distance transport of Rn under favorable conditions (Mogro-Campero and Fleischer, 1977).

Kristiansson and Malmqvist (1982) first proposed the carrier gas mechanism to explain the transport of Rn. The authors conducted measurements at a Pb mine field and then compared the results with theoretical calculations of the transport rate. Their results revealed that diffusion alone could not explain the long-distance Rn transport; in other words, an ascending stream of gas is

required to explain the migration of Rn. However, Rn concentration is never high enough for it to form a gas stream by itself, so that a carrier gas is required. The authors termed this process the carrier gas transport mechanism. It infers that a stream of a carrier gas, forms bubbles in the water-filled cracks or fissures (two domains, water and gas domains co-exist) and carries Rn. The carrier gas moves upward with a velocity much higher than the diffusion rate. This hypothesis was confirmed by additional laboratory experiments and field work (Etiopie and Lombardi, 1995, 1996; Yang et al., 2003b; Ciotoli et al., 2005). The carrier gas mechanism was established and largely applied in the study of Rn;  $\text{CO}_2$  was considered as the most probable carrier gas for soil Rn (Etiopie and Lombardi, 1995; Guerra and Lombardi, 2001; Baubron et al., 2002; Ciotoli et al., 2004).

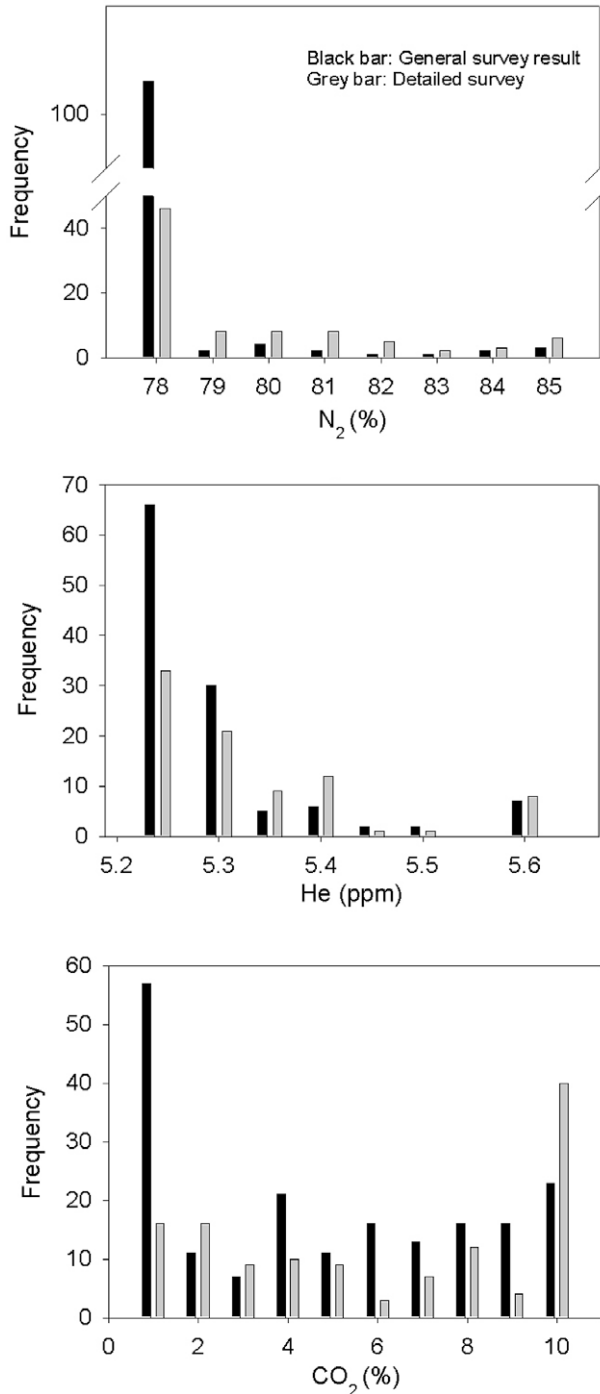
Although it is well accepted that  $\text{CO}_2$  might be the carrier gas for Rn, the mechanism for He migration in the solid earth is still a problem. The possibility of diffusional transport of He within the crust had been ruled out by previous studies (Torgersen and Clarke, 1985; Torgersen, 1989). The possibility of advective transport is also small due to the fact that the He flux is variable and episodic (Torgersen and Clarke, 1992). Moreover, the distinct decoupled nature between the He and Ar flux is difficult to explain by advection (Torgersen et al., 1989). Torgersen (1980) provided a detailed model explaining how He could enter from the solid phase into the liquid phase. In his model, the liquid phase could help He to be transported to the atmosphere. The question then would be which fluid, groundwater or geogas, is the carrier to help He migrate. In this study, field observations are provided to suggest the possible mechanism of He migration in the studied area. Although  $\text{N}_2$  is thought to be the most probable carrier gas for He, which could serve as supporting evidence for the carrier gas mechanism, the possibility of a groundwater transport mechanism is not totally dismissed.

## 2.2. Methodology

A 1.5 m long hollow stainless steel tube, 3 cm in diameter, was used for sampling soil gas. A metal tip was added at the front of the tube in order to penetrate the ground easily. About 1.2 m of the tube was hammered into ground. A thinner but longer pipe was

**Table 1**  
Statistics of gas composition for all the collected samples.

	Mean	Standard deviation	Min	Max	Sampling points
<i>General survey along Hsincheng Fault</i>					
CO <sub>2</sub>	2.9%	2.8	0%	13.8%	118
He	5.28 ppm	0.13	5.15 ppm	6 ppm	118
N <sub>2</sub>	76.5%	3.4	74.1%	91.2%	118
<i>Detailed survey inside Hsinchu Science Park (HSP)</i>					
CO <sub>2</sub>	4.6%	4.1	0%	17.2%	85
He	5.32 ppm	0.15	5.15 ppm	6.07 ppm	85
N <sub>2</sub>	79.1%	6.0	73.7%	97.5%	85



**Fig. 2.** Distribution of N<sub>2</sub>, He and CO<sub>2</sub> concentrations in samples collected during general and detailed surveys. The x-axis shows the concentration for each gas species; the y-axis shows the number of sample in each concentration range.

**Table 2**

Gas compositions of samples for carbon, nitrogen and helium isotopic analysis.

Sample no.	Ar (%)	N <sub>2</sub> (%)	O <sub>2</sub> (%)	CO <sub>2</sub> (%)	[ <sup>4</sup> He] (ppm)	N <sub>2</sub> /Ar	Excess <sup>b</sup> N <sub>2</sub> (%)
21-1	0.93	87.07	5.62	6.38	5.40	93.3	9.0
21-2	1.03	90.00	3.05	5.92	5.46	85.1 <sup>a</sup>	4.0
21-3	1.08	93.74	5.18	n.d.	5.68	87.2	3.8
21-4	1.57	85.14	13.29	n.d.	5.37	54.1	-46
21-5	1.39	85.10	9.79	3.73	5.33	61.2	-31
21-6	0.96	83.45	12.38	3.22	5.33	87.4	3.6
21-7	1.01	85.03	12.00	1.96	5.33	84.2	0.6
21-8	0.99	81.84	17.17	n.d.	5.32	83.1	-0.5
21-9	0.98	82.76	9.21	7.06	5.30	84.8	1.2
21-10	0.95	79.36	18.82	0.88	5.24	83.9	0.3
21-11	0.98	79.41	17.12	2.49	5.24	80.7	-2.8
21-12	1.02	82.91	12.95	3.13	5.28	81.6	-2.1
21-13	0.98	81.01	14.22	3.78	5.24	82.3	-1.3
21-14	0.92	79.57	19.51	n.d.	5.32	86.5	2.7
21-15	1.03	87.20	8.72	3.05	5.40	84.6	1.0
23-1	0.97	81.52	13.14	4.38	5.48	84.4	0.8
23-2	0.98	82.63	12.41	3.98	5.24	84.3	0.7
23-3	1.02	88.93	8.34	1.71	5.08	87.1	3.6
23-4	1.06	89.61	8.16	1.17	n.d.	84.8	1.3
23-5	0.98	88.12	5.92	4.97	5.40	89.6	5.9
23-6	1.01	83.44	10.86	4.70	5.79	82.7	-0.9

n.d.: not detectable.

<sup>a</sup> N<sub>2</sub>/Ar ratio of this sample was determined at the Ocean Research Institute of the University of Tokyo.

<sup>b</sup> Excess N<sub>2</sub> is defined as [N<sub>2</sub>]<sub>sample</sub> - [Ar]<sub>sample</sub> \* (78.082/0.934).

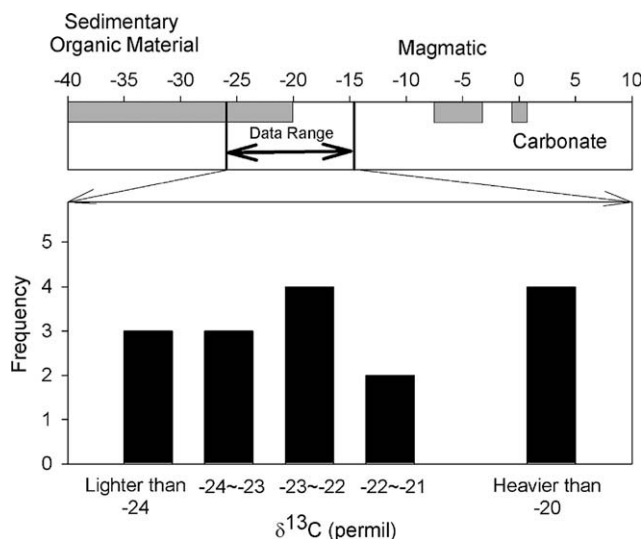
then inserted into the hollow tube in order to separate the tip from the hollow tube. The hollow tube was connected to an already designed pipe line and a hand pump or automatic pump; after flushing, the soil gas could be pumped out and stored in vacuum sampling bags or bottles for further analyses.

Soil gas surveys along 11 profiles transecting the Hsincheng Fault, one of the active faults in NW Taiwan, were conducted (Fig. 1a and b). During the 4-month survey, a total number of 118 samples were collected. In addition, 85 samples were collected in a very confined area in the Hsinchu Science Park (HSP) (rectangle in Fig. 1c) in order to determine suitable places for setting up monitoring stations. The distance interval of sampling points is several hundreds to several tens of meters depending on local conditions and the resolution required. For weekly monitoring, 11 temporary monitoring sites (site-1 to -11) were set up along a profile across the Hsincheng Fault inside or near the HSP (Fig. 1c). This profile is about 1.5 km long, and the average distance between each sampling point from site-1 to -9 is ca. 50 m. Sites -10 and -11 were set up as background control points, so they were outside the HSP and away from the fault trace. Site-1 was very close to a trench which was drilled for a paleoseismologic study of the Hsincheng Fault (Chen et al., 2003). Soil gas samples were collected every week from each station for 2 months inside the HSP in order to continuously monitor the concentration variation of several gas species (He, CO<sub>2</sub>, N<sub>2</sub>, Ar + O<sub>2</sub>). Gas compositions of these samples

**Table 3**  
Carbon, nitrogen, and helium isotopic results.

Sample no.	$\Delta^4\text{He}$ (%)	Ra	$[\text{He}/^{20}\text{Ne}]$	$\delta^{13}\text{C}$ (‰)	$\delta^{15}\text{N}$ (‰)	$F^a$
21-1	11.6	$1.03 \pm 0.03$	0.33			
21-2	-0.8	$1.01 \pm 0.06$	0.30	-21.87	$-0.30 \pm 0.01$	73
21-3	11.9	$0.96 \pm 0.03$	0.19			
21-4	-1.1	$1.05 \pm 0.06$	0.28			
21-5	17.4	$1.01 \pm 0.03$	0.34	-22.59		75
21-6	14.2	$1.00 \pm 0.04$	0.38	-23.13		77
21-7	1.1	$1.04 \pm 0.06$	0.30	-22.69		76
21-8	-0.7	$1.02 \pm 0.06$	0.32			
21-9	1.5	$0.95 \pm 0.04$	0.35	-22.39		75
21-10	3.4	$0.98 \pm 0.04$	0.32	-14.56		44
21-11	-7.9	$1.06 \pm 0.06$	0.31			
21-12	12.6	$1.05 \pm 0.04$	0.29	-24.18		81
21-13	-1.6	$1.01 \pm 0.05$	0.31	-23.58		79
21-14	4.7	$1.04 \pm 0.05$	0.33	-23.96		81
21-15	7.0	$1.07 \pm 0.05$	0.33	-22.31		74
23-1	4.6	$0.97 \pm 0.06$	0.32	-18.92		61
23-2	0.0	$0.96 \pm 0.05$	0.33	-16.72		53
23-3	-3.1	$0.94 \pm 0.05$	0.31	-21.71		72
23-4	49.3	$1.04 \pm 0.06$	0.29	-18.43		59
23-5	3.1	$0.96 \pm 0.04$	0.31	-25.87		88
23-6	10.4	$1.04 \pm 0.04$	0.34	-24.87		84

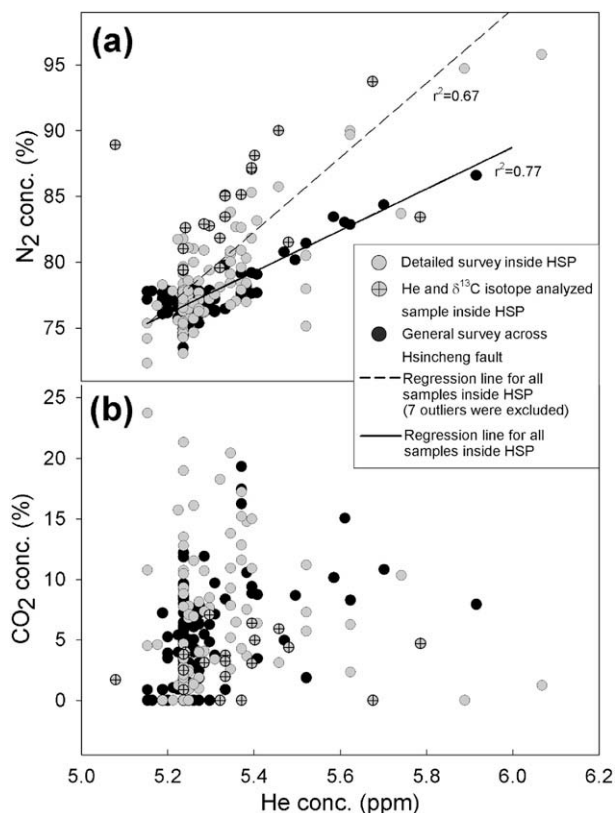
<sup>a</sup>  $F$  denotes the fraction of mixing between  $\text{CO}_2$  from limestone and biogenic processes. End member isotopic value was set to be 0‰ for  $\text{CO}_2$  from carbonate and -30‰ which is the mid-point in the range for a biogenic source in Hoefs (2004).  $F$  is calculated following:  $F * (-30\text{‰}) + (1 - F) * (0\text{‰}) = \delta^{13}\text{C}_{\text{sample}}$ .



**Fig. 3.**  $\delta^{13}\text{C}$  values of the soil  $\text{CO}_2$  in this study. Grey bars show the ranges for the major components of terrestrial  $\text{CO}_2$  sources, based on Hoefs (2004). For  $\text{CO}_2$  from carbonate, the  $\delta^{13}\text{C}$  is  $\sim 0\text{‰}$ . For a magmatic source, the range is ca.  $-3.4\text{‰}$  to  $-7.8\text{‰}$ . For sedimentary organic material, the range is  $-20\text{‰}$  to  $-40\text{‰}$ . The results show that most soil  $\text{CO}_2$  is from sedimentary organic material with little input from carbonate derived  $\text{CO}_2$  (gas from a magmatic source is excluded by the He isotopic results; details discussed in text).

were analyzed immediately after collection, by micro gas chromatography (VARIAN CP4900) (for Ar +  $\text{O}_2$ ,  $\text{N}_2$  and  $\text{CO}_2$ ; the peaks for  $\text{O}_2$  and Ar overlapped each other, so their individual concentration could not be differentiated), and He leak detector (ASM100HDS, Alcatel). Details of analytical procedures and errors were described in Fu et al. (2005) and Walia et al. (2010).

In order to determine He and C isotopes, an additional 21 samples were collected inside the HSP. The gas compositions of these samples were determined by another gas chromatograph (GC, SRI 8610C) with two thermal conductivity detectors (TCD) and one flame ionic detector (FID) in order to differentiate Ar and  $\text{O}_2$ . For

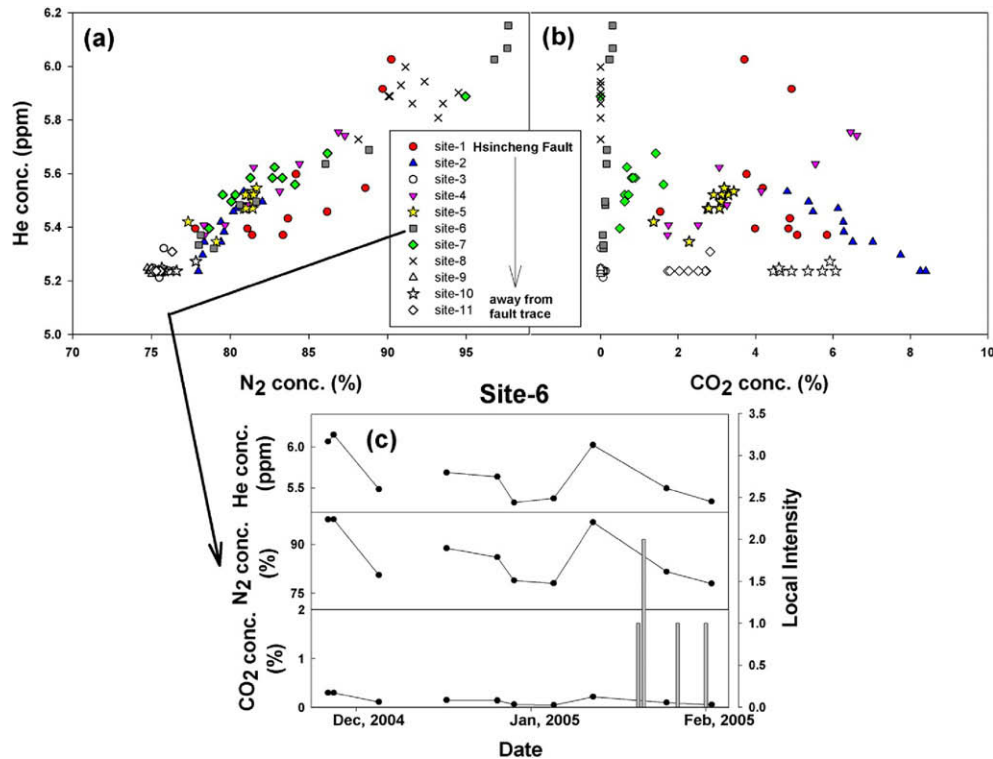


**Fig. 4.** Correlation between  $\text{N}_2$ , He and  $\text{CO}_2$  concentration. Grey dots are samples collected during the detailed survey; black dots are samples collected during the general survey. All symbols with a cross inside indicate samples with He and  $\delta^{13}\text{C}$  isotopic analysis. (a) Good correlation can be observed between He and  $\text{N}_2$ . Two proportional correlations were observed indicating that such relationships may vary due to different spatial scales. (b) No significant correlation was observed between He and  $\text{CO}_2$  concentrations.

a more detailed description of the analytical procedure has been described by Lee et al. (2005). A noble gas mass spectrometer (Micromass 5400) was used for the analysis of He isotopic ratios and Ne/He ratios. The total error is less than 2.5% including analytical error for samples and working standards and also long-term variations of standards (Yang et al. 2005a, 2006a). Since the soil gas samples are largely affected by atmospheric air, the traditional expression of He isotopic ratio ( $^3\text{He}/^4\text{He}$ ) cannot clearly show the variability in order to determine the origin of the samples. Therefore, results are expressed as the concentration of excess  $^4\text{He}$  ( $\Delta^4\text{He}$ ) which is normalized to atmospheric air concentration. Excess  $^4\text{He}$  can be expressed as

$$\Delta^4\text{He}(\%) = \frac{([\text{He}]_{\text{in sample}} - [\text{He}]_{\text{in air}})}{[\text{He}]_{\text{in air}}} \times 100 \quad (1)$$

For  $\delta^{13}\text{C}_{\text{CO}_2}$  analysis,  $\text{CO}_2$  from soil gas samples was trapped by liquid  $\text{N}_2$  in a vacuum gas purification system. Then, the purified  $\text{CO}_2$  was sealed in a glass bead for further isotopic analyses using a Finnigan MAT mass spectrometer. One of the samples was sent to the Ocean Research Institute of the University of Tokyo for  $\delta^{15}\text{N}_{\text{N}_2}$  and  $\text{N}_2/\text{Ar}$  ratio analysis. Analytical errors for  $\delta^{15}\text{N}_{\text{N}_2}$  and  $\text{N}_2/\text{Ar}$  were about 0.3‰ and 3‰, respectively (Takahata et al., 1998). Carbon and N isotopic ratios are expressed relative to Pee Dee Belemnite for C and atmospheric air for N. In order to calculate the excess  $\text{N}_2$  concentration relative to atmospheric air, Ar concentration was used as a reference for calculation because Ar concentration is relatively constant compared to  $\text{O}_2$  and  $\text{N}_2$ . In addition, Ar is less affected by the processes discussed here. The calculation is done according to the formula below:



**Fig. 5.** Weekly monitoring results of 11 monitoring sites. Each symbol indicates an individual site. Points of a symbol refer to samples collected at different dates. (a) Good spatial and temporal correlation could be observed between N<sub>2</sub> and He in most sites (sites -1, -2, -4, -5, -6, -7, -8); poor correlation occurred at sites that are far away from the fault trace (sites -9, -10, -11). This may suggest that distance from the fault trace may also control gas emission. (b) A significant proportional relationship between CO<sub>2</sub> and He occurred only at sites -4 and -5. Most sites do not exhibit significant correlations (or even inverse correlation at site-2). (c) The variation of He, N<sub>2</sub>, CO<sub>2</sub> concentration, and local seismic intensity with time in site-6 is plotted as an example showing the temporal variation of different gases and relationship with seismic events.

$$[\text{N}_2 \text{ conc.}]_{\text{excess}} = [\text{N}_2 \text{ conc.}]_{\text{sample}} - [\text{Ar conc.}]_{\text{sample}} \times 78.082/0.934 \quad (2)$$

where 78.082 and 0.934 are the percentages of N<sub>2</sub> and Ar for theoretical atmospheric air.

### 3. Results and discussion

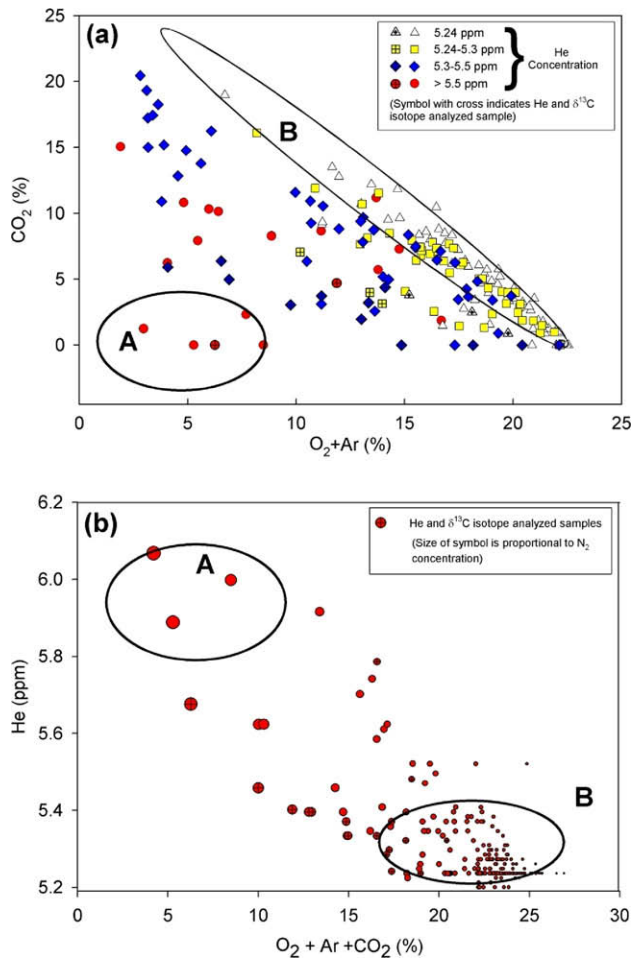
The soil gas survey conducted in this study can be divided into two parts. One is the general survey involving 118 samples, taken along 11 profiles across the Hsincheng Fault; the other is the detailed survey which was focused in the HSP for the purpose of establishing monitoring stations. For the detailed survey, a total of 85 samples were collected within several hundred m<sup>2</sup> inside the HSP. All of these 203 samples were analyzed for N<sub>2</sub>, CO<sub>2</sub>, Ar + O<sub>2</sub>, and He concentration; CH<sub>4</sub> was only found in three samples, so its concentration variation will not be discussed. The means as well as minimum and maximum compositional values are given in Table 1 and Fig. 2. In addition to those 203 samples, another 21 samples were collected for He,  $\delta^{13}\text{C}_{\text{CO}_2}$  and  $\delta^{15}\text{N}_{\text{N}_2}$  isotopic analysis (Tables 2 and 3). The excess <sup>4</sup>He ranges from -7.9 to 49.3. Values of  $\delta^{13}\text{C}_{\text{CO}_2}$  fall in the range of -25.9‰ to -14.6‰. These  $\delta^{13}\text{C}_{\text{CO}_2}$  values show that CO<sub>2</sub> gas in the study area is dominated by biogenic CO<sub>2</sub> with a small admixture of CO<sub>2</sub> from carbonate (Fig. 3). The  $\delta^{15}\text{N}_{\text{N}_2}$  value and N<sub>2</sub>/Ar ratio of sample #21-2 are -0.3‰ and 85.1 (Tables 2 and 3), respectively, indicating that the source of N<sub>2</sub> gas is dominated by atmospheric air (Littke et al., 1995). The N<sub>2</sub>/Ar ratios of other samples fall within the range of 54.1–93.3 (Table 2) which also indicate the atmospheric-origin of N<sub>2</sub> in the soil gas of this region.

Several lines of evidence, which will be discussed in the following paragraphs, suggest that N<sub>2</sub> is the most probable carrier gas for

He emission in the studied area. First, good spatial and temporal correlations are observed for N<sub>2</sub> and He, but do not exist for CO<sub>2</sub> and He. Second, from isotope and gas composition data, CO<sub>2</sub> in the studied area behaves differently from He, suggesting that they are not derived from the same source. A two-source mixing model is proposed to explain the soil gas system and its relationship to the fault system in this area.

Soil N<sub>2</sub> concentrations show a relatively good correlation with He concentration ( $r^2 = 0.67$  for the detailed survey and  $r^2 = 0.77$  for the general survey), while such a relationship is not observed between CO<sub>2</sub> and He ( $r^2 = 0.11$ ) (Fig. 4a and b), in this study. In Fig. 4a, two proportional relationships could be observed. One is pertaining to the general survey along the 11 profiles across the Hsincheng Fault (black points in Fig. 4a); the other is derived from the detailed survey in the HSP (grey points in Fig. 4a). Two different proportional relationships confirm that such relationship does exist in general but varies due to different sampling locations or resolution. However, such relationships cannot be observed between He and CO<sub>2</sub> (Fig. 4b).

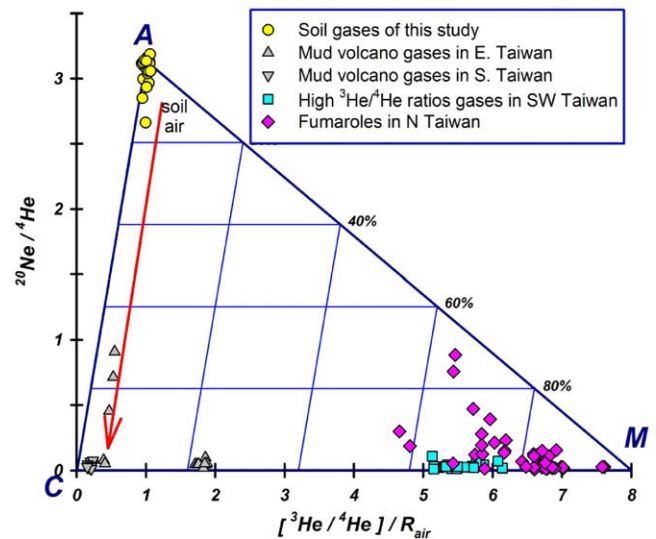
The spatial and temporal relationship between N<sub>2</sub>, CO<sub>2</sub> and He is shown in Fig. 5. In Fig. 5a and b, each symbol represents a monitoring station; points with different symbols represent samples collected on different days at that station. The temporal variations are illustrated for site-6 in Fig. 5c as an example. In Fig. 5a, 6 out of 11 stations show good proportional correlation between N<sub>2</sub> and He concentration indicating that their concentrations have the same temporal variation at more than half of the stations. Moreover, the distance away from the fault trace might be one of the factors controlling the soil gas He concentrations (and N<sub>2</sub>) in the studied area. Systematic variation of the He concentration can be observed from site-1 to -7 (excluding site-3 which might be affected by wet soil); however, such variations could not be observed in samples



**Fig. 6.** Scatter plot of the concentrations of various gas species. All symbols with a cross inside indicate samples with He and δ<sup>13</sup>C isotopic analysis. (a) All samples can be classified in four categories according to their He concentration. Lower CO<sub>2</sub> and O<sub>2</sub> + Ar concentration would be expected as He concentration increases. (b) The correlation between He concentration and the sum of Ar, O<sub>2</sub> and CO<sub>2</sub> indicating the inverse relationship. The size of the symbols is proportional to the N<sub>2</sub> concentration. Two sources could be identified from the figure. Source A is the source with high He and N<sub>2</sub>-content while both CO<sub>2</sub> and O<sub>2</sub> contents are low. Source B is the source with a constant concentration sum of O<sub>2</sub>, Ar, and CO<sub>2</sub> while N<sub>2</sub> and He concentrations are close to the atmospheric air value. The CO<sub>2</sub> and O<sub>2</sub> content in source B would follow an inverse proportional relationship indicating variable degrees of atmospheric air content and/or microbial activity involvement.

collected from site-8 to -11. As shown in Fig. 5c, the He and N<sub>2</sub> concentration of site-6, as an example, have very similar temporal variation. On the other hand, such a variation is not clear between CO<sub>2</sub> and He at this site. Local intensity of seismic events during that time period is also shown in the figure indicating that some relationship between seismic activities and He and N<sub>2</sub> concentration may exist. However, the temporal resolution of the sampling is insufficient to identify the relationship at this stage. A long term continuous monitoring station has been set up in order to verify the real cause of the gas variations (Walia et al., 2009).

In Fig. 5b, a poor correlation is observed between CO<sub>2</sub> and He concentrations. Some sites do show a proportional correlation, for examples, sites -4 and -5. However, other sites exhibit inverse proportional relationships (site-2) or no significant relationship (sites -1, -3, -6, -7, -8, -9, -10 and -11). In particular, very different behavior between CO<sub>2</sub> and He was observed for some sites. High CO<sub>2</sub> but low He concentrations were observed at sites -9 and -10; whereas low CO<sub>2</sub> but high He concentrations were observed at sites -6 and -8, suggesting that degassing of CO<sub>2</sub> and He are re-



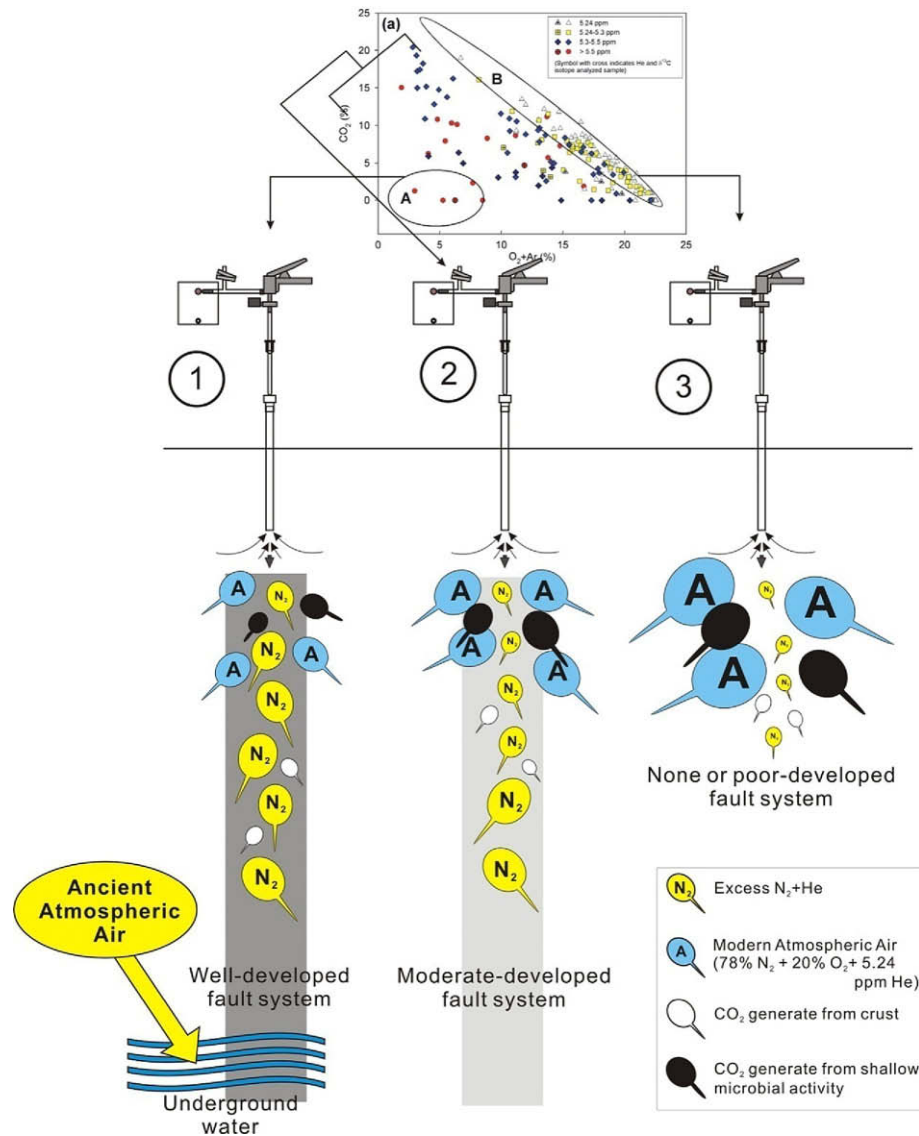
**Fig. 7.** Three-component plot of He and Ne isotopes for soil gas samples in this study. Gas samples from mud volcanoes in southern Taiwan and fumarole samples from northern Taiwan are also shown for comparison (data sources: Yang et al., 2003a, b; Lan et al., 2007; Lee et al., 2008). A: air; C: crust; M: mantle component.

lated to different mechanisms. In summary thus, the above discussions show that good spatial and temporal relationships exist between N<sub>2</sub> and He but not between CO<sub>2</sub> and He. Such relationships indicate that N<sub>2</sub> is the most plausible carrier gas for He in this area.

The relationship between CO<sub>2</sub>, Ar + O<sub>2</sub>, and He concentration is shown in Fig. 6a. If only samples with a He concentration equal to 5.24 ppm (triangles in Fig. 6a) are considered, their CO<sub>2</sub> and Ar + O<sub>2</sub> concentration show an inverse proportional relationship. This might represent the differing contents of air. Another explanation is that O<sub>2</sub> derived from air was consumed by biogenic processes producing CO<sub>2</sub> (Rixon and Bridge, 1968; Ross and Roberts, 1970; Gregory and Durrance, 1985; Lovell, 2000); thus, this relationship may represent the involvement of biogenic processes. No matter which process occurred in the studied area, they both indicate that the gas is derived from shallow sources.

If the samples with He concentration >5.24 ppm are further considered, a progressive trend could be observed in that He concentration increases as both CO<sub>2</sub> and O<sub>2</sub> concentrations decrease (Fig. 6a). This trend is further demonstrated in Fig. 6b where He concentration was plotted versus the sum of CO<sub>2</sub> and O<sub>2</sub> concentrations. As CO<sub>2</sub> and O<sub>2</sub> concentration decrease, both N<sub>2</sub> and He concentration increase, but it must be noted that if a gas is removed from a gas mixture, the concentration of remaining components increases. The increase of N<sub>2</sub> concentration could probably be a consequence of decreasing CO<sub>2</sub> and O<sub>2</sub> concentration. However, in samples with high He concentration, N<sub>2</sub> is the only gas left implying that N<sub>2</sub> is the only possible carrier for He. The amount of excess N<sub>2</sub> may be calculated on the basis of the Ar content in the soil gas samples if it is assumed all of the Ar is from air. As shown in Table 2, up to 9% of excess N<sub>2</sub> was obtained indicating that not all of the N<sub>2</sub> in the soil gas is from modern air. Another source with high N<sub>2</sub> concentration is required to explain the excess gas. So far, it has been shown that the degassing behavior of He and CO<sub>2</sub> is very different, so that it is difficult to argue for CO<sub>2</sub> as a unique phase to carry He. Moreover, a source with high N<sub>2</sub> concentration is required. Such a high N<sub>2</sub>-content source might also be characterized by high He-content based on the proportional relationship between N<sub>2</sub> and He (Fig. 4a).

The observation that He and CO<sub>2</sub> might be derived from different sources could be further verified by their isotopic



**Fig. 8.** Sketch of the model explaining the gas mixtures and migration mechanisms in the study area. Three groups of gas behaviors were distinguished. (1) A well-developed fault system serves as a pathway for excess N<sub>2</sub> and He for migrating upward. The excess N<sub>2</sub> observed might be derived from ancient atmospheric air dissolved in groundwater. Very little admixture from source B is observed. (2) Where only a moderately-developed fault system exists, less excess N<sub>2</sub> and He migrate upward. Some degree of admixture from source B is observed. (3) In poorly-developed fault systems soil gas is dominated by atmospheric or biogenic gas.

compositions. Based on the He isotopic results (Fig. 7), the soil gases are dominated by an air component with some crustal input. Gas from a magmatic source could be ruled out; otherwise, some signal should be observed in the He isotopic data. Based on the  $\delta^{13}\text{C}_{\text{CO}_2}$  value ( $-25.9 \sim -14.6\text{‰}$ ), CO<sub>2</sub> is considered to be dominated by a biogenic source ( $\sim 70\text{--}80\%$ ), which is impossible for He generation. Regarding the source of excess N<sub>2</sub> in the studied area, the  $\delta^{15}\text{N}_{\text{N}_2}$  value ( $-0.3\text{‰}$ ) and N<sub>2</sub>/Ar ratio (85.1) (Table 3), both indicate an atmospheric air source of N<sub>2</sub>. This implies that the excess N<sub>2</sub> might be derived from ancient atmospheric air which dissolved in groundwater. During the residence in the aquifer, most O<sub>2</sub> in ancient atmospheric air was consumed due to its higher activity. As a result, while the groundwater is flowing across a fault zone, excess N<sub>2</sub> carrying He can be released from the groundwater and enter into surrounding soil gas. This could also indicate that the carrier gas mechanism is capable of explaining the transport of He in the studied area.

A model is proposed to explain the soil gas system in the studied area. Two sources of soil gases in this region could be defined

from Fig. 6a and b. Source A is characterized by (1) excess He and N<sub>2</sub>, (2) none or very low CO<sub>2</sub>, O<sub>2</sub> and Ar contents; (3) CO<sub>2</sub> and O<sub>2</sub> concentration do not follow the inverse proportional relationship observed in source B. The characteristics of source B are: (1) He and N<sub>2</sub> concentrations are close to atmospheric air values; (2) CO<sub>2</sub> and Ar + O<sub>2</sub> concentration follows an inverse proportional relationship. From these characteristics and the isotopic characteristics, it is believed that source B is a mixture of atmospheric air and biogenic gas, whereas source A is considered as abiogenic gas ultimately derived from ancient atmospheric air that dissolved in underground water. During the transport of groundwater, most O<sub>2</sub> was consumed by some microbial activities or reactions with host rocks. The dissolved ancient air contributes the excess N<sub>2</sub> in source A. Where the fault system is well-developed, the gas from source A might carry He and migrate upward.

Considering the two-source model and that distance from the fault trace may affect gas behavior, the gas samples were classified into three groups (Fig. 8). Group 1 is where gas from source A migrates rapidly through a well-developed fault system; during the

migration there is none or very little influence from other sources. Group 1 could correspond to those samples with He concentration >5.5 ppm (squares in Fig. 6a). Group 2 represents gas samples that are mixtures of sources A and B which could correspond to samples with He concentrations >5.5 ppm but <5.24 ppm (diamonds and squares in Fig. 6a). Samples in this group represent gas migrating through an only moderately-developed fault system, so that it suffers a certain degree of mixture with source B. Group 3, which could correlate to samples with He concentrations equal to 5.24 ppm (triangles in Fig. 6a), represents the condition where no or only a poorly-developed fault system exist. Because there is no pathway for gas from source A to migrate, He and N<sub>2</sub> concentration is close to the value of atmospheric air.

#### 4. Conclusions

- (1) Significant correlations are observed in this study between soil gas concentrations of N<sub>2</sub> and He. Such a relationship is not observed between CO<sub>2</sub> and He.
- (2) The monitoring results show that the distance from the fault trace might be one factor controlling the variations of the He concentration suggesting that the fault system could provides pathways for gas to migrate upward from deep sources.
- (3) The isotopic composition of the gases enables identification of the origins of soil CO<sub>2</sub>, He and N<sub>2</sub>: CO<sub>2</sub> is dominated by biogenic sources with little input from carbonate sources, and He is dominated by atmospheric gas with some input from the crust; N<sub>2</sub> is dominantly derived from ancient atmospheric air.
- (4) Nitrogen is enriched by up to 9% relative to air suggesting that not all of the N<sub>2</sub> observed in soil gas is derived from modern atmospheric air; some might be from ancient atmospheric air. Ancient atmospheric air that dissolved in underground water contributes the excess N<sub>2</sub> and is the most probable carrier gas for He in this area.
- (5) It is proposed that the carrier gas mechanism is the most plausible mechanism to explain the transport of He in this study and that N<sub>2</sub> is the most probable carrier gas. A two-source mixing model is proposed to explain the soil gas system in this area. Source A is characterized by high He and N<sub>2</sub> and low CO<sub>2</sub> and O<sub>2</sub> contents. In source B, CO<sub>2</sub> and O<sub>2</sub> contents follow an inverse proportional relationship indicating different degrees of air content or involvement of microbial activities. Helium and N<sub>2</sub>-contents in source B are close to the value of the atmospheric air.

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